

The background of the slide is a blue-tinted photograph of a mountain range. In the foreground, there is a dense forest of evergreen trees. The mountains in the background are layered, with the closest ones appearing darker and more detailed, and the ones further away becoming progressively lighter and more hazy. The sky is a clear, light blue.

# **Description of the Model of Emissions of Gases and Aerosols from Nature (MEGAN)**

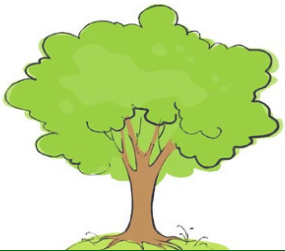
# Overview

## Background

- ◆ What is MEGAN? Briefly
- ◆ Why model BVOCs?

## Practical

- ◆ How MEGAN works
- ◆ MEGAN in CLM
- ◆ Applications



# Overview

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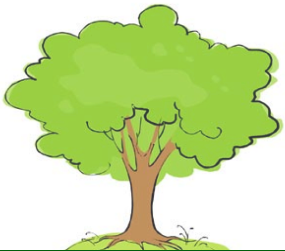
## Practical

- ◆ How MEGAN works
- ◆ MEGAN in CLM
- ◆ Applications

### More Details:

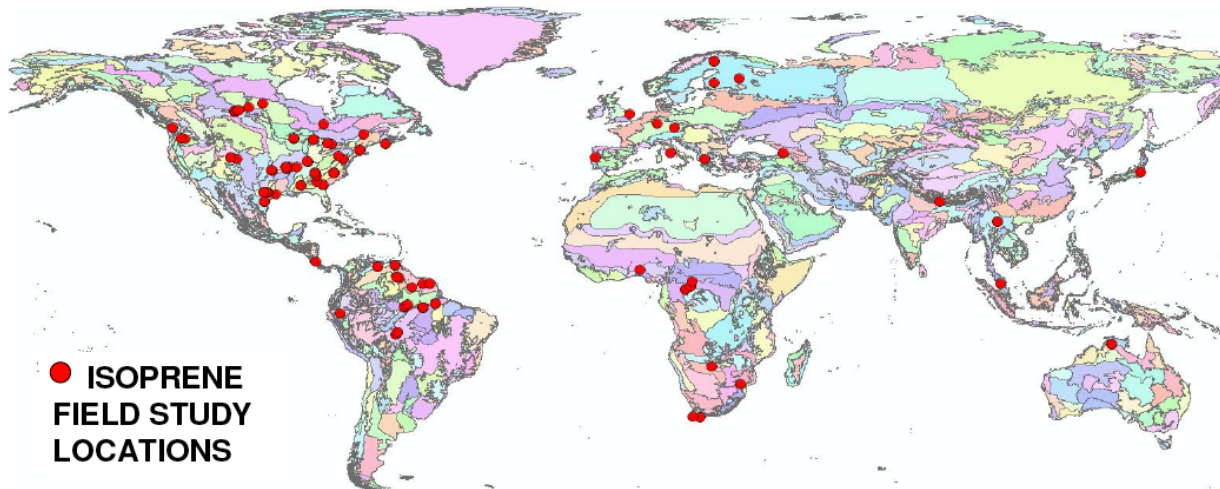
Guenther et al. 2012: MEGAN2.1, Geosci. Model Dev.  
[doi:10.5194/gmdd-5-1503-2012](https://doi.org/10.5194/gmdd-5-1503-2012)

Guenther et al. 2006: MEGAN 1.0, Atmos. Chem. Phys.,  
[doi:10.5194/acp-6-3181-2006](https://doi.org/10.5194/acp-6-3181-2006)



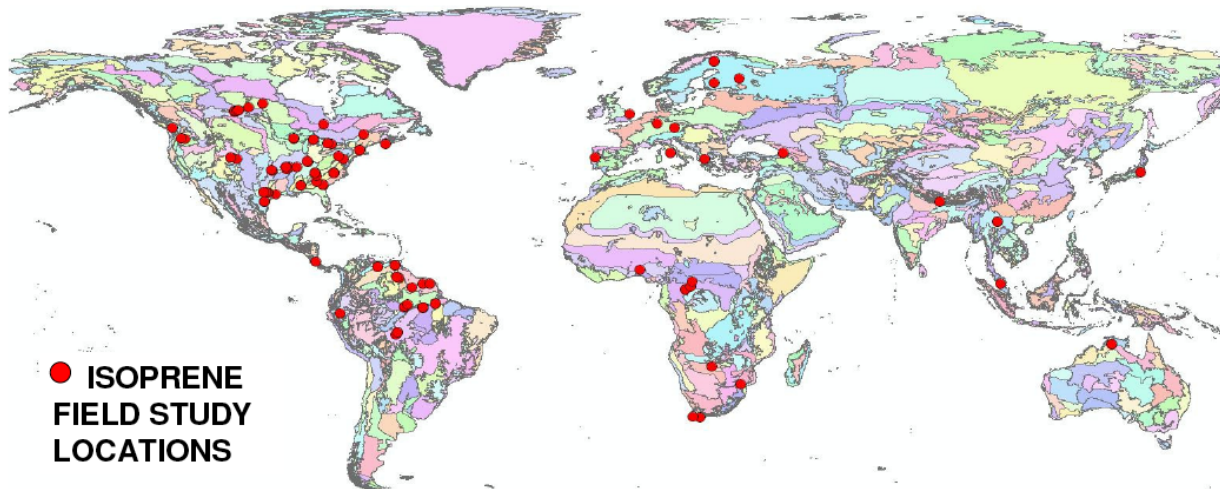
# What is MEGAN?

- **MEGAN = Model of Emissions of Gases and Aerosols from Nature**
- **Estimates *NET* biogenic volatile organic compound (BVOC) emissions to the atmosphere**
- **Uses >80 enclosure studies and >90 field measurement studies to parameterize what controls emission rates**



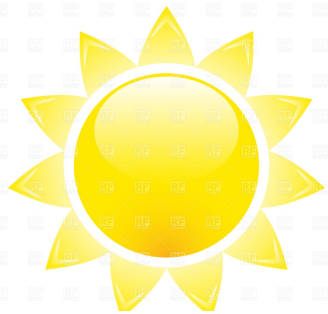
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*Why attempt to model BVOCs?*

# Why Model BVOC Emissions?



+



+



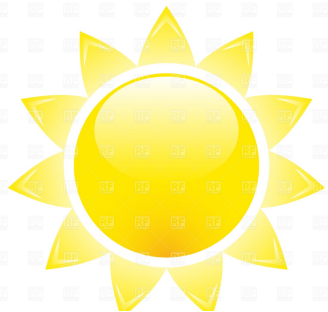
= **Ozone**



# Why Model BVOC Emissions?

Sunlight + Volatile Organics + Nitrogen Oxides = Ozone

$h\nu$  + VOCs +  $\text{NO}_y$  =  $\text{O}_3$



+



+

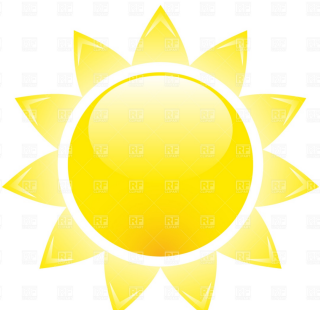


= Ozone

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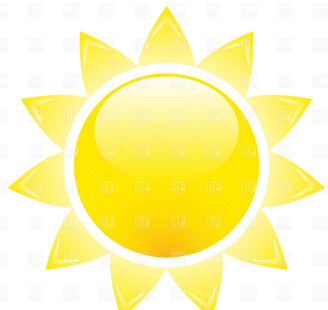


# Why Model BVOC Emissions?



Sunlight + Volatile Organics + Nitrogen Oxides = Ozone

hv + VOCs + NO<sub>y</sub> = O<sub>3</sub>



+



+

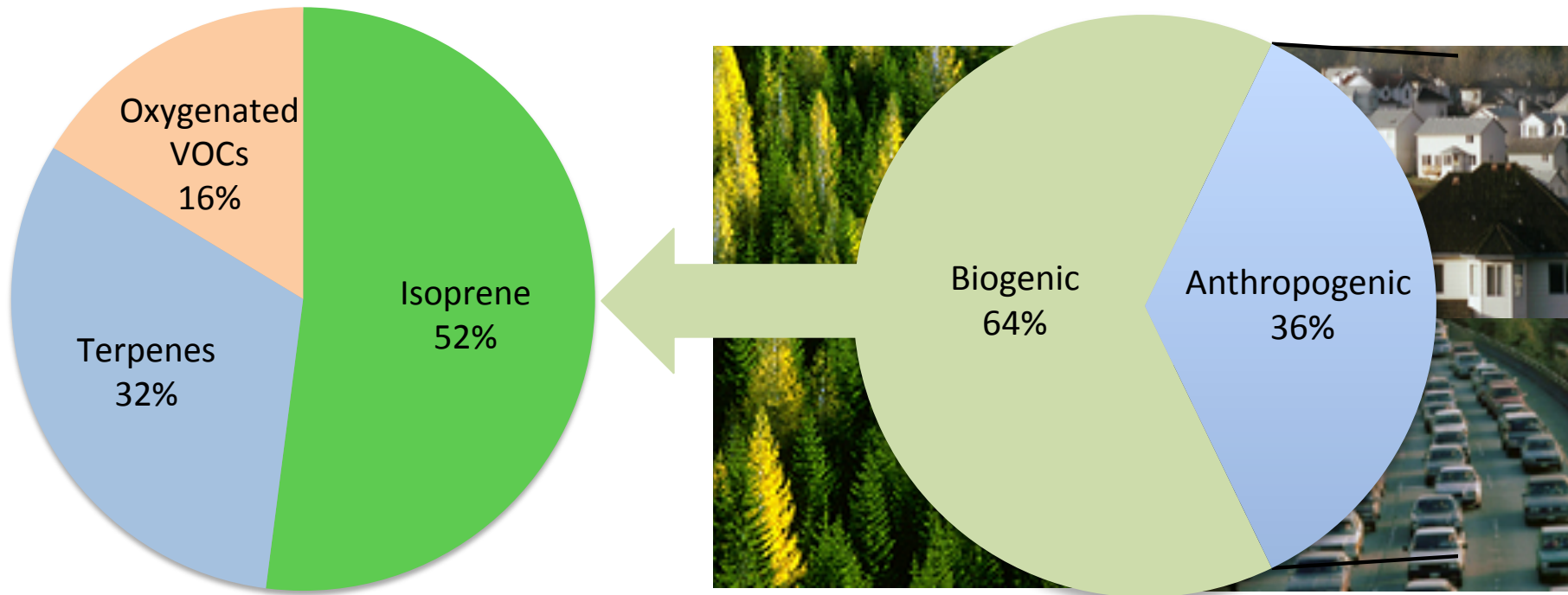


= Ozone

# Sources of BVOCs

- The majority of VOCs come from natural (biogenic) sources
- Isoprene accounts for a majority of VOC reactivity

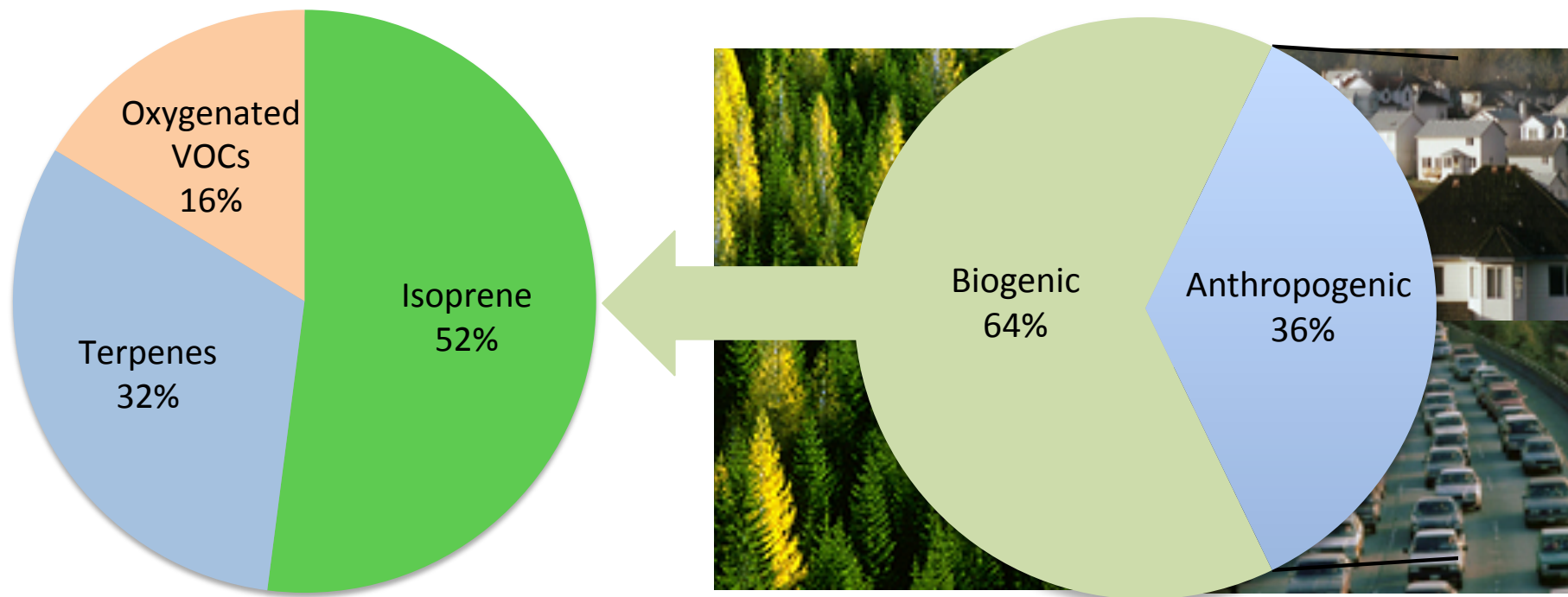
## Percentages for US



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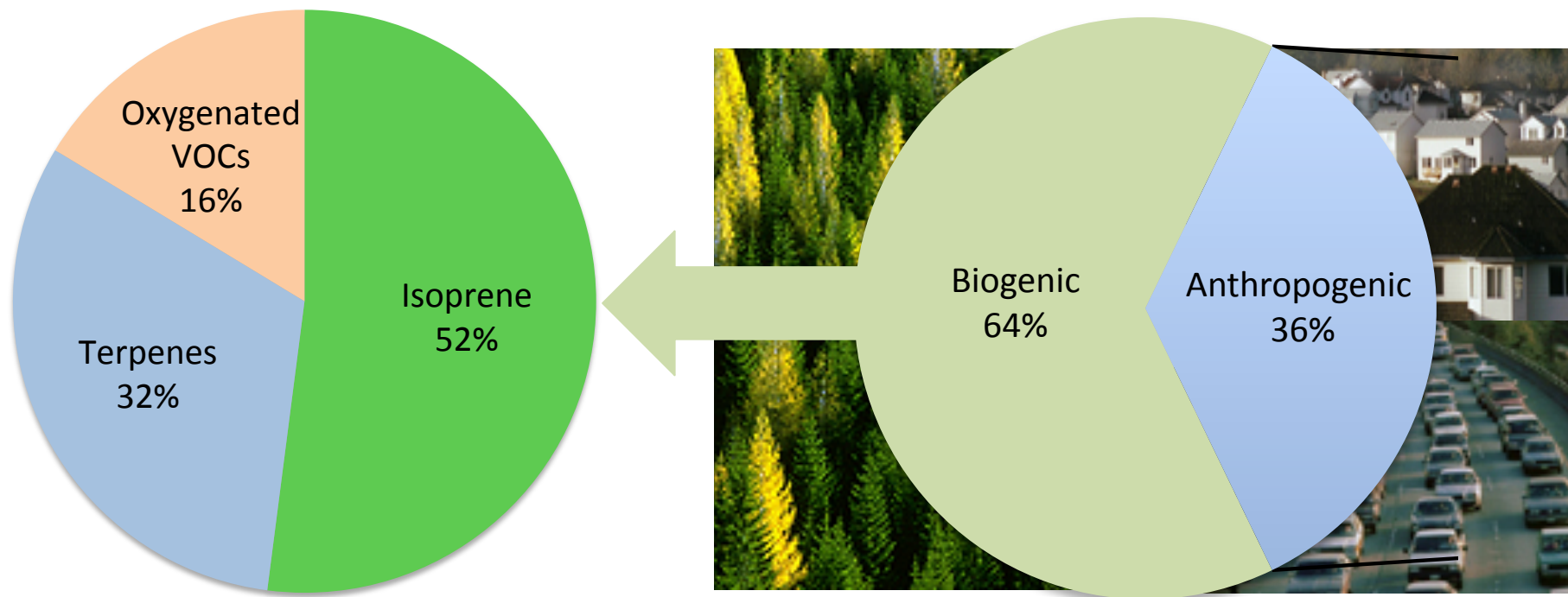


**\*\* Biogenic VOC Emission highly depends on vegetation type and meteorology**

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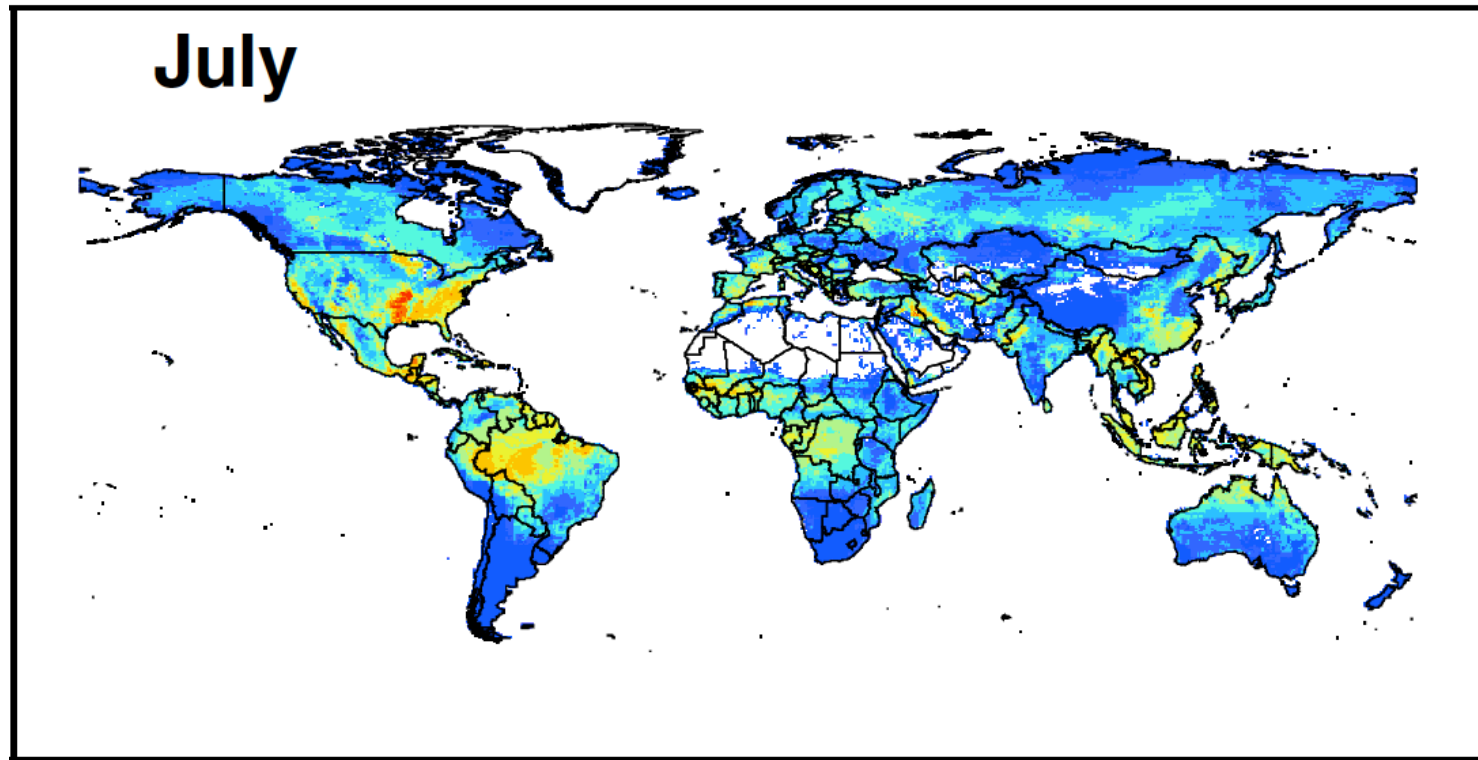
## Percentages for US



**\*\* Biogenic VOC Emission highly depends on vegetation type and meteorology**

# Global Contributions of BVOCs

- 1-2% of carbon assimilated by the terrestrial ecosystem per year is emitted back to the atmosphere as BVOCs globally
- Can be much higher regionally under stressed conditions (~10%)



# Biogenic VOC Classes

- There are several BVOC classes:

- Isoprene



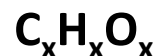
- Monoterpenes

 $2^*isoprene$ 

- Sesquiterpenes

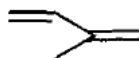
 $3^*isoprene$ 

- Oxygenated VOCs

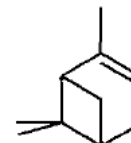


- BVOC classes have similar oxidation pathways in the atmosphere

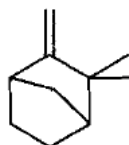
Isoprene



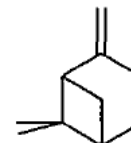
$\alpha$ -Pinene



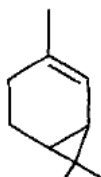
Camphene



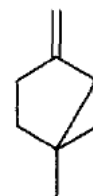
$\beta$ -Pinene



2-Carene



Sabinene





# Why Plants Emit VOCs

## Plant protection against stress



Thermotolerance



Oxidative stress tolerance

Photoprotection

## Plant defense

Indirect defense against herbivores

Direct defense against pathogens

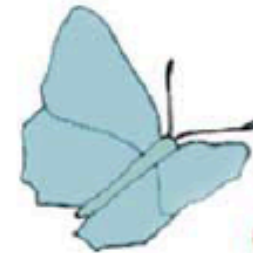
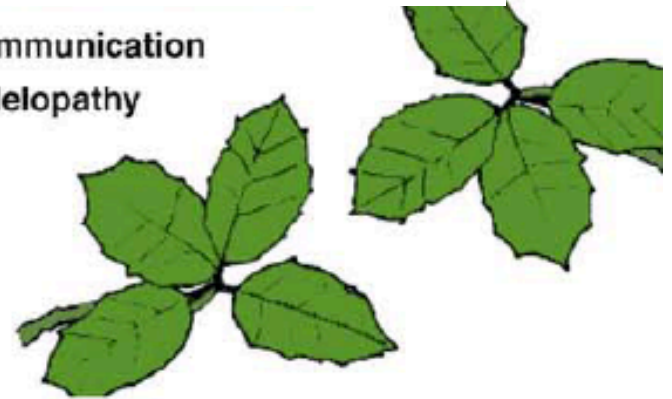
Direct defense against herbivores



## Plant-Plant interaction

Communication

Allelopathy



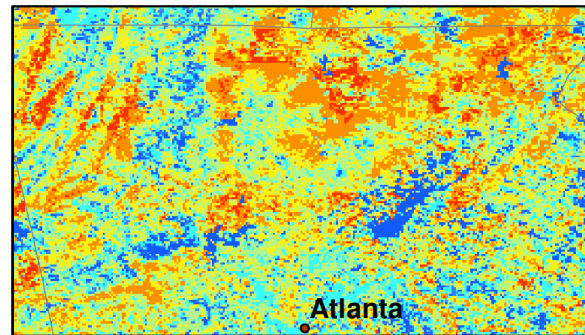
## Plant Reproduction

Pollination

Fruit and Seed dispersal

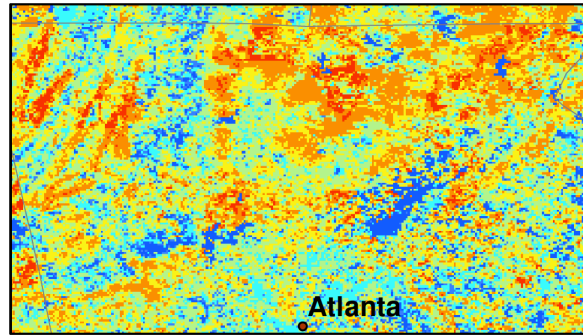
# How MEGAN Models Emissions?

Modify a Standard Emission Factor ( $\epsilon$ )



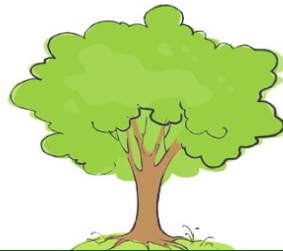
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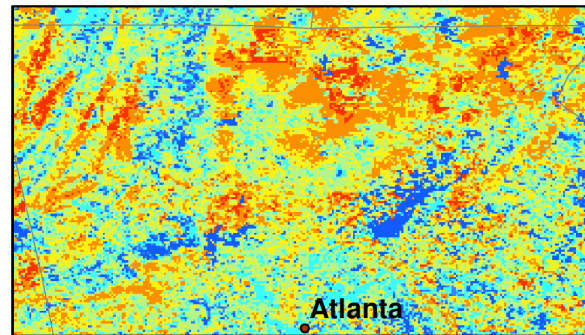
$\epsilon$  = emission rate for a specific environment condition

This standard emission factor is then modified by actual conditions to estimate an emission rate



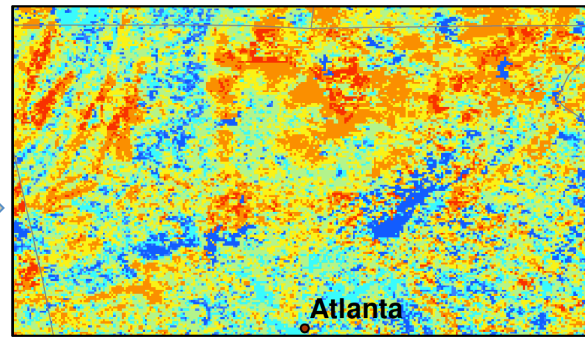
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# How MEGAN Models Emissions?

Modify a Standard Emission Factor ( $\epsilon$ )



Leaf Temperature

CO<sub>2</sub>

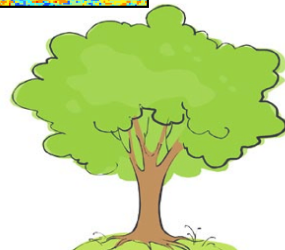
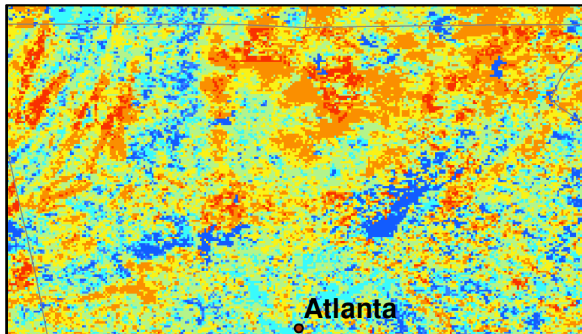


Soil Moisture

# How MEGAN Models Emissions?

Two ways of prescribing a Standard Emission Factor ( $\epsilon$ )

## Species-Specific Map



## By PFT

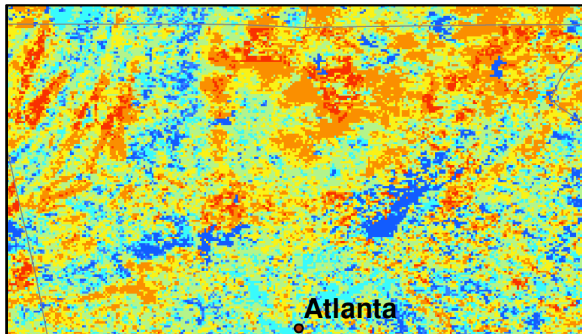
CLM PFT Number	Description	Land Area	Isoprene Tg yr <sup>-1</sup>	MT Tg yr <sup>-1</sup>	Other Tg yr <sup>-1</sup>
	Bare	40.7			
1	Needleleaf Evergreen Temperate Tree	5.46	1.61	7.38	13.2
2	Needleleaf Evergreen Boreal Tree	10.6	5.9	6.63	9.52
3	Needleleaf Deciduous Boreal Tree	1.46	0.0002	0.52	0.89
4	Broadleaf Evergreen Tropical Tree	15.6	244	82.9	127
5	Broadleaf Evergreen Temperate Tree	2.64	21.9	4.0	8.71
6	Broadleaf Deciduous Tropical Tree	12.9	178	45.0	74.3
7	Broadleaf Deciduous Temperate Tree	5.33	35.4	5.86	13.1
8	Broadleaf Deciduous Boreal Tree	2.14	4.79	0.99	2.02
9	Broadleaf Evergreen Temperate Shrub	0.18	0.23	0.08	0.33
10	Broadleaf Deciduous Temperate Shrub	4.15	21.8	6.77	16.4
11	Broadleaf Deciduous Boreal Shrub	9.33	2.93	1.07	3.3
12	Arctic C3 Grass	4.94	0.97	0.02	1.45
13	Cool C3 Grass	14.3	11.2	0.25	26.1
14	Warm C4 Grass	13.2	5.93	0.49	51.3
15	Crop1	16.3	0.02	0.36	44.5
Total (all PFTs)		159	535	162	390



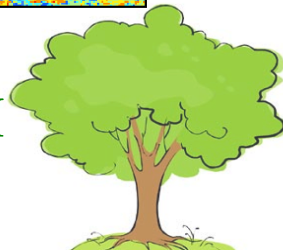
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Two ways of prescribing a Standard Emission Factor ( $\epsilon$ )

## Species-Specific Map



**\*\* Only for  
Isoprene in CLM**



## By PFT

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# MEGAN Algorithm

$$\text{Emission Flux} = \varepsilon * \text{LAI} * C_E * \gamma_T * \gamma_{\text{PAR}} * \gamma_{\text{age}} * \gamma_{\text{SM}} * \gamma_{\text{CO2}} * \rho$$

Input

MEGAN Calculation

\*\* Each Gamma Factor ( $\gamma$ ) has a response curve derived from observations

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## Input

Photosyn. Active Radiation

Leaf Temperature

CO<sub>2</sub>

Soil Moisture

Leaf Area Index

Standard Emission Factor

Consistency Factor

In-canopy Reactivity

## MEGAN Calculation

$\gamma_{\text{PAR}}$

$\gamma_T$

$\gamma_{\text{CO}_2}$

$\gamma_{\text{SM}}$

LAI

$\varepsilon$

$C_E$

$\rho$

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Leaf Temperature

$\gamma_T$

CO<sub>2</sub>

$\gamma_{\text{CO}_2}$

~~Soil Moisture~~

$\gamma_{\text{SM}}$

= 1

Leaf Area Index

LAI

Standard Emission Factor

$\varepsilon$

Consistency Factor

$C_E$

~~In-canopy Reactivity~~

$\rho$

= 1

**\*\* Each Gamma Factor ( $\gamma$ ) has a response curve derived from observations**

# List of BVOCs Estimated by MEGAN

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## Compound Class

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Isoprene

Myrcene

Sabinene

Limonene

3-Carene

*t*- $\beta$ -Ocimene

$\beta$ -Pinene

$\alpha$ -Pinene

Other Monoterpenes

$\alpha$ -Farnesene

$\beta$ -Caryophyllene

Other Sesquiterpenes

232-MBO

Methanol

Acetone

CO

Bidirectional VOC

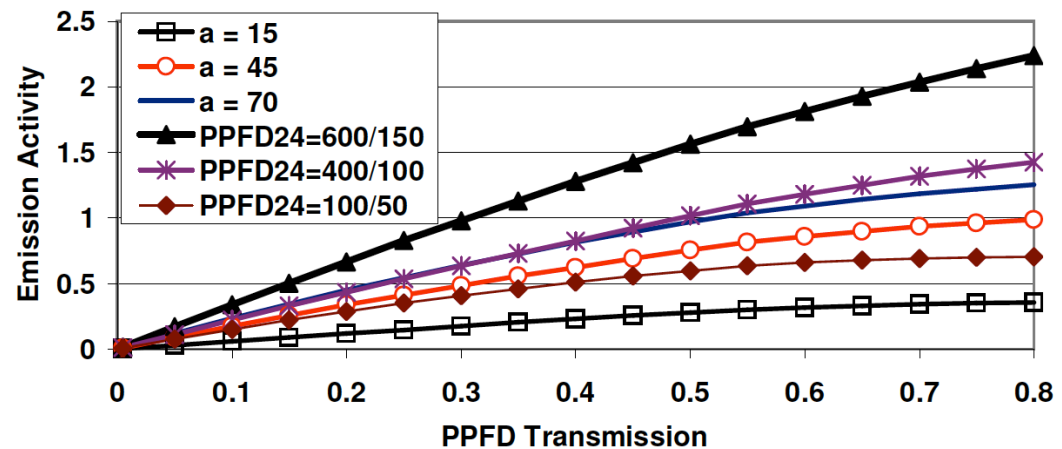
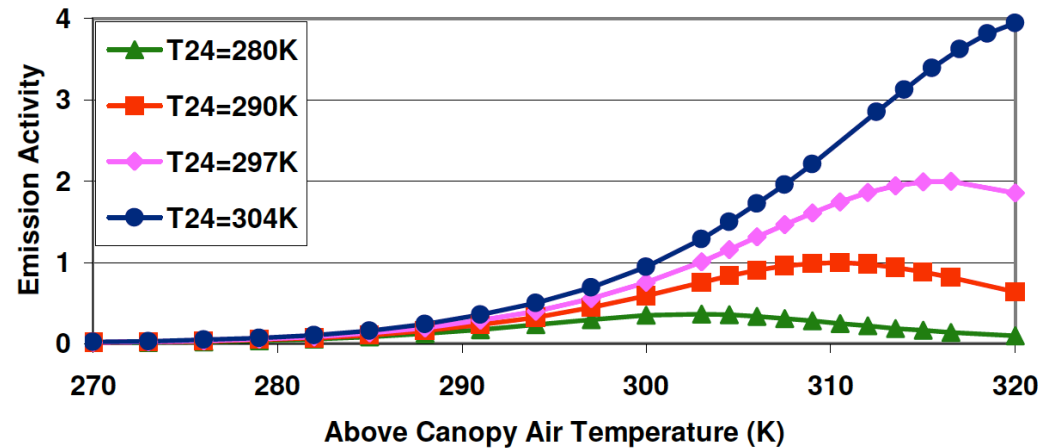
Stress VOC

Other VOC

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# Sensitivity to Different Gamma Factors

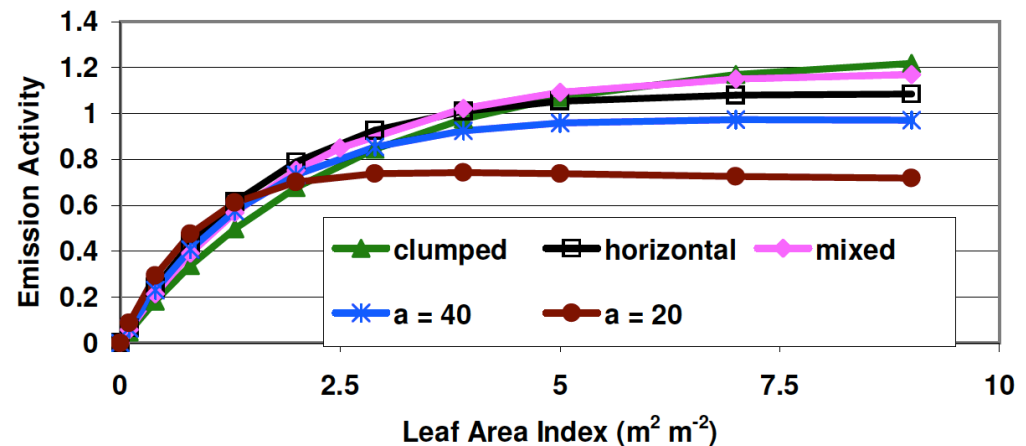
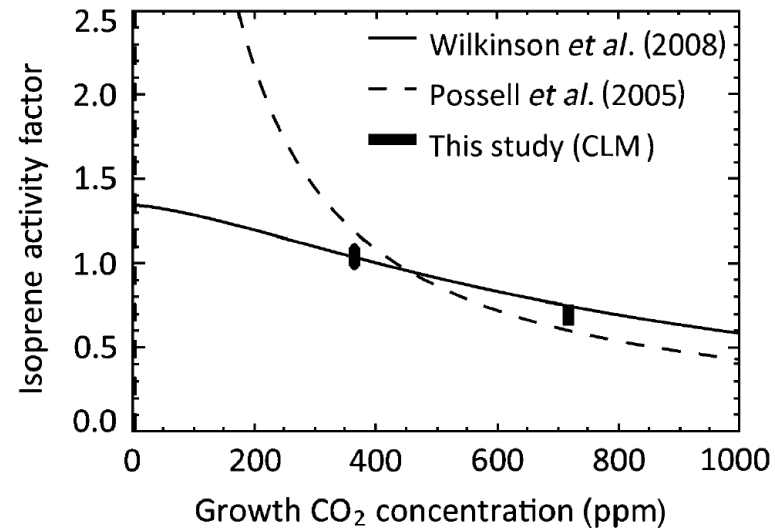
- Emissions increase rapidly with temperature
- $\gamma_T$  also captures past 24 and 240 hour effects
- PAR response is more gradual
- Also captures past PAR conditions
- $\gamma_T$  and  $\gamma_p$  account for shaded and sunlit canopy fraction





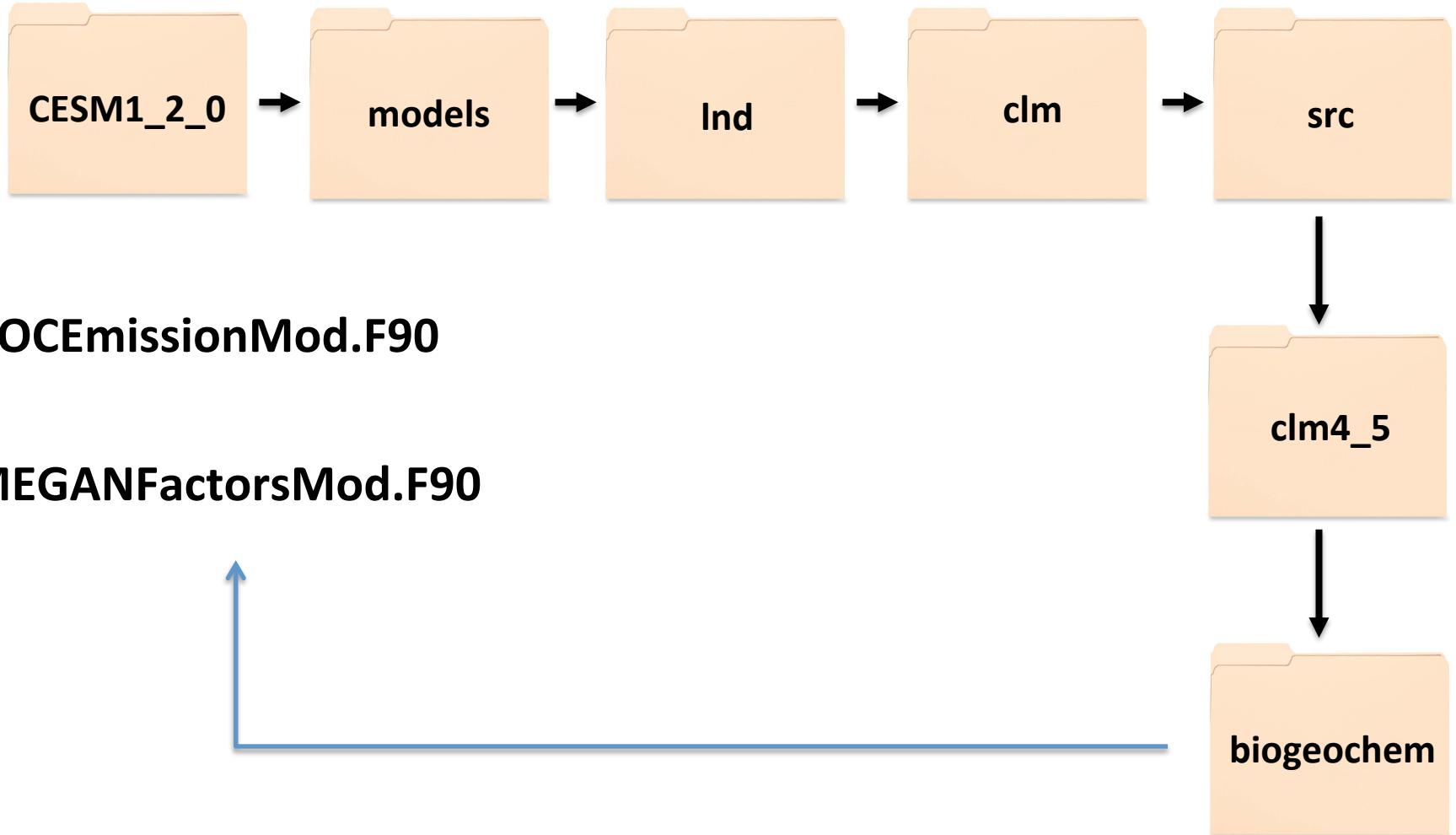
# Sensitivity to Different Gamma Factors

- **CO<sub>2</sub> emissions suppress BVOC emissions**
- **Only included for isoprene; =1 for all other**
- **More LAI equals more emissions to a point**
- **Note that activity factor is rarely above 1**



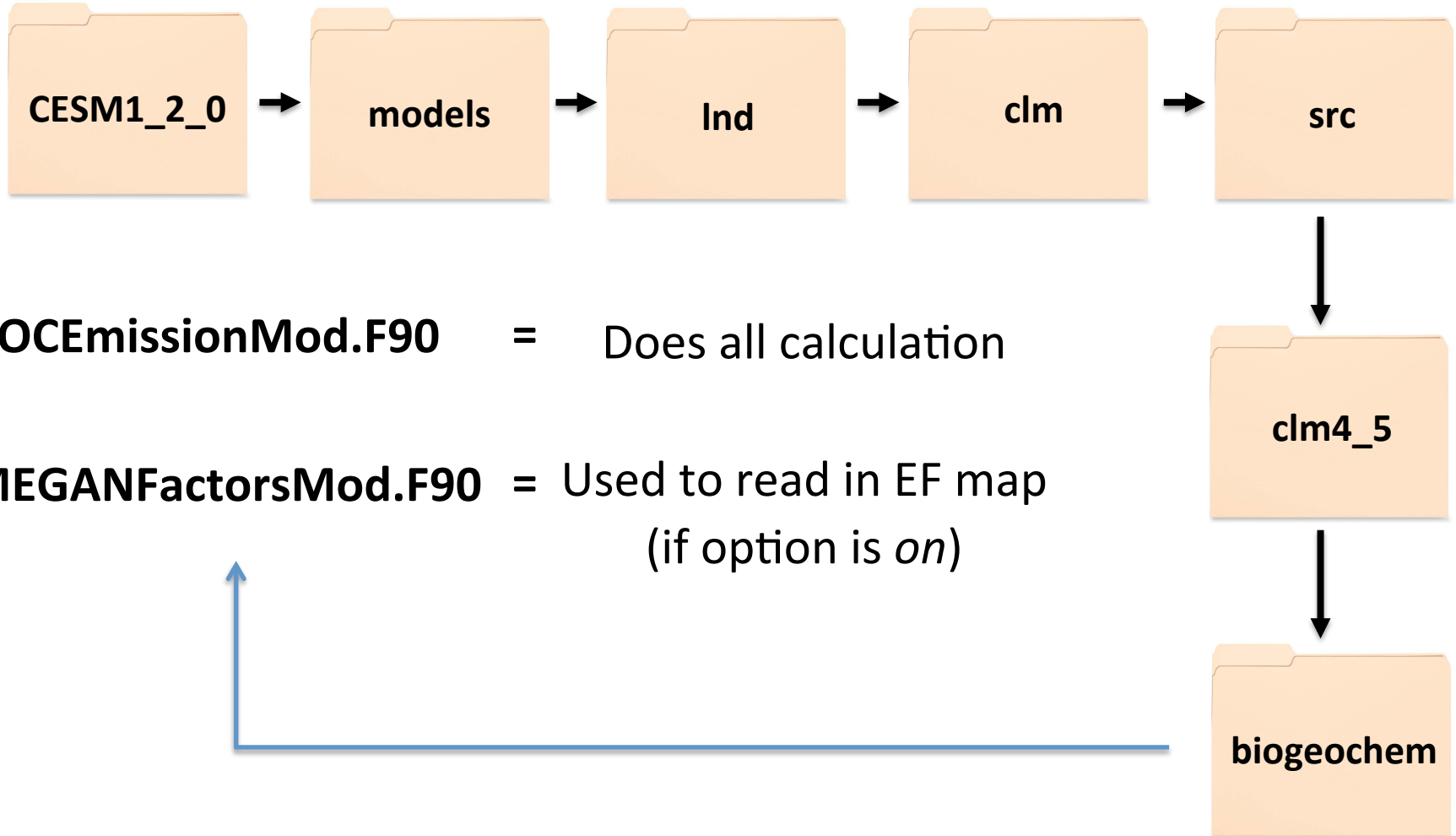
# Where does MEGAN live in the CESM?

cesm1\_2\_0/models/Ind/clm/src/clm4\_5/biogeochem



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cesm1\_2\_0/models/Ind/clm/src/clm4\_5/biogeochem



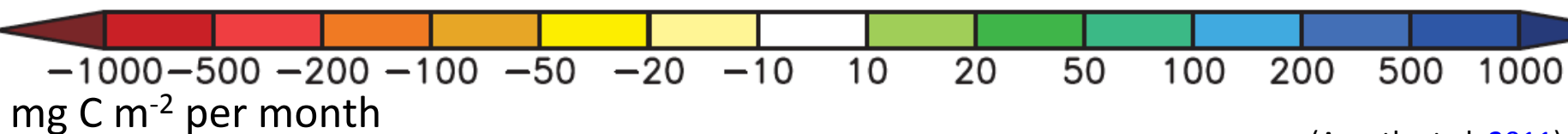
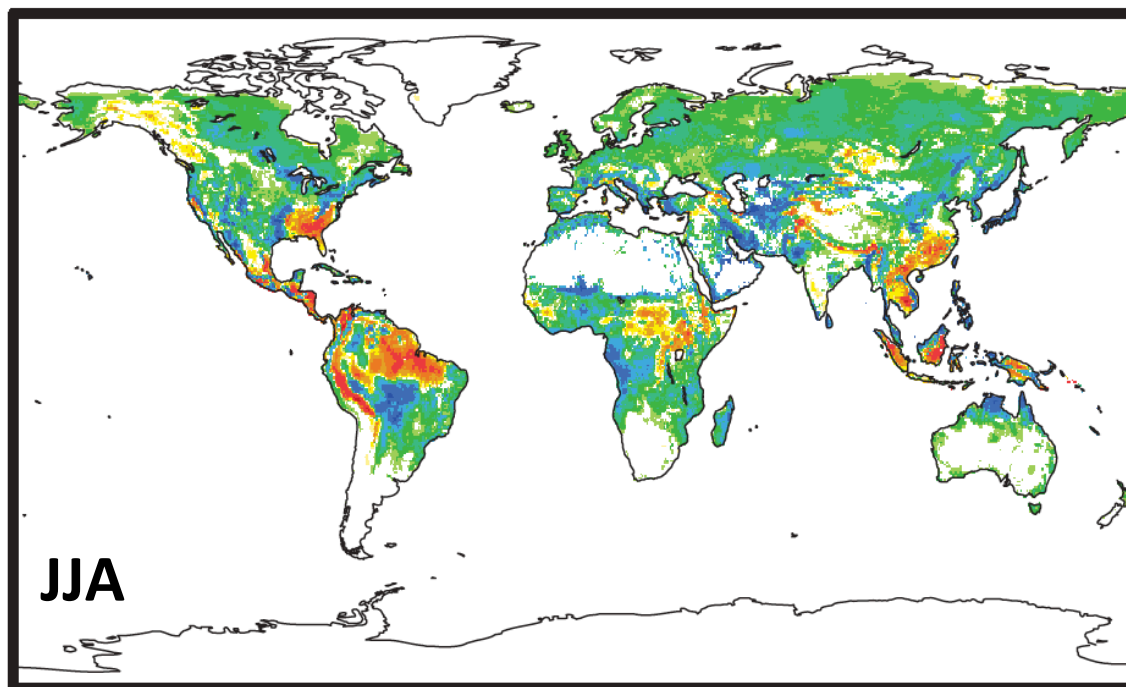
# MEGAN Workflow in VOC Emission Mod

1. Initialize variables
2. Read EF map (for isoprene only) if user wants otherwise use PFT
3. Loop over all vegetation types
  - Calculate all the gamma factors based CLM forcing
  - Scale the standard EF by the gamma factor to get emissions
4. Send to CAM/Chemistry model if running coupled

# Sensitive to Forcing Data

Only changing the meteorological forcing dataset

**NCEP minus CRU**



(Arneeth et al. [2011](#))

# Applications of MEGAN

- **Estimate global or regional carbon “lost” to the atmosphere**
- **Implement with CAM and atmospheric chemistry to understand impacts of BVOCs on**
  - **Aerosol-cloud feedbacks**
  - **Ozone air quality**
  - **Direct radiative forcing**
- **Influence of changing land cover type**
- **Emissions changes under different future projections**

# Summary of MEGAN

