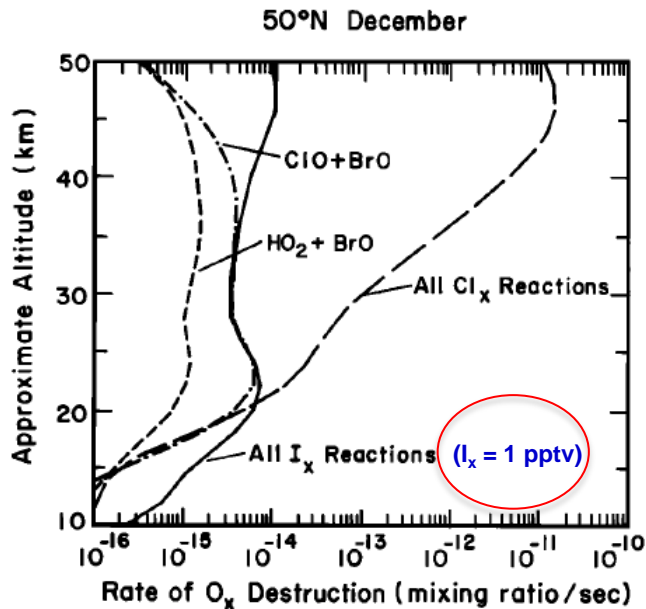


Global modelling of upper tropospheric-lower stratospheric iodine: budget and implications for ozone

Alfonso Saiz-Lopez, Carlos Ordóñez, Juan Carlos Gomez Martin
(Laboratory for Atmospheric and Climate Science, CIAC-CSIC,
Toledo, Spain)

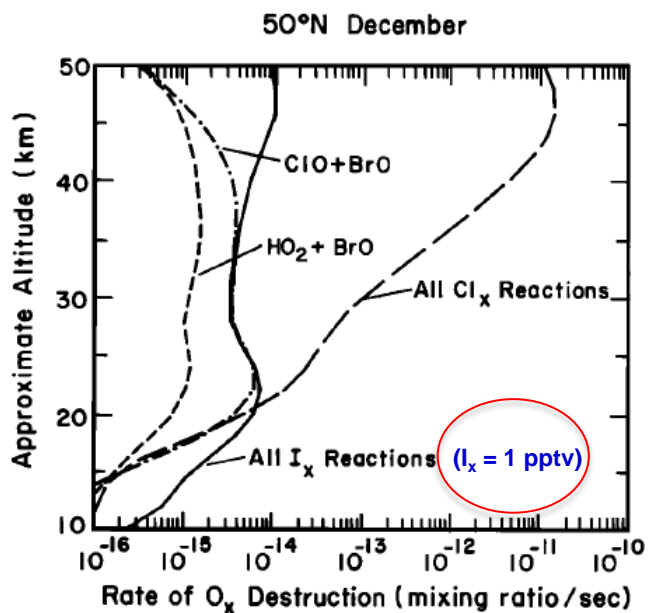
Jean-François Lamarque, Douglas E. Kinnison, Simon Tilmes
(NCAR, Boulder, CO, USA)

“On the role of iodine in ozone depletion”, Solomon et al., JGR, 1994



Solomon et al.,: “iodine loadings of 1 pptv and rapid inter-halogen photochemical reactions involving iodine could be of substantial importance to ozone loss in the lowermost stratosphere”.

Solomon et al., JGR, 1994



Solomon et al., JGR, 1994

Since then:

- New kinetic information on iodine available
- Attempts to detect reactive iodine in the UTLS

Wennberg et al., JGR, 1997

Pundt et al., JAC, 1998

Wittrock et al., GRL, 2000

Berthet et al., JGR, 2003

Bösch et al., JGR, 2003

Butz et al., ACP, 2009



Most recent analysis: < 0.1 pptv IO, OIO in lower stratosphere in northern high and mid-latitudes, and tropics

Estimated total inorganic iodine: ~ 0.2 pptv I_y
(Photochemical 1-D model)

- 1) Budget and partitioning of iodine species in the global UTLS using CCM
- 2) Re-evaluation of suggested role of iodine in UTLS O₃ destruction
- 3) Relative contribution of iodine to halogen-driven UTLS O₃ destruction

CAM-Chem (Lamarque et al., *Geosci. Model Dev.* 2012):

- Global Chemistry-Climate Model
- 1.9° (lat) x 2.5° (lon) horizontal resolution
- 26 vertical levels (surface to ~ 4 hPa)
- time step = 30 min

Tropospheric Halogen Chemistry

- Implementation of VSL (lifetime < 6 months) halogenated sources from the ocean.
- Emissions following Chl-a over tropics
- Catalytic release from sea-salt
- Photochemistry
- Dry / wet deposition
- 160 species, 427 reactions

VSL halogen sources in CAM-Chem

Source gas	Local Lifetime (WMO, 2010)	Main loss	
CH ₂ BrCl	137 days	OH, hv	
CH ₂ Br ₂	123 days	OH, hv	
CHBrCl ₂	78 days	OH, hv	
CHBr ₂ Cl	59 days	hv, OH	
CHBr ₃	24 days	hv, OH	
CH ₃ I	7 days	hv, OH	(Bell et al., 2002)
CH ₂ ICl	~ 2–3 h	hv	
CH ₂ IBr	~ 1 h	hv	
CH ₂ I ₂	~ 5 min	hv	
I ₂	~ secs	hv	

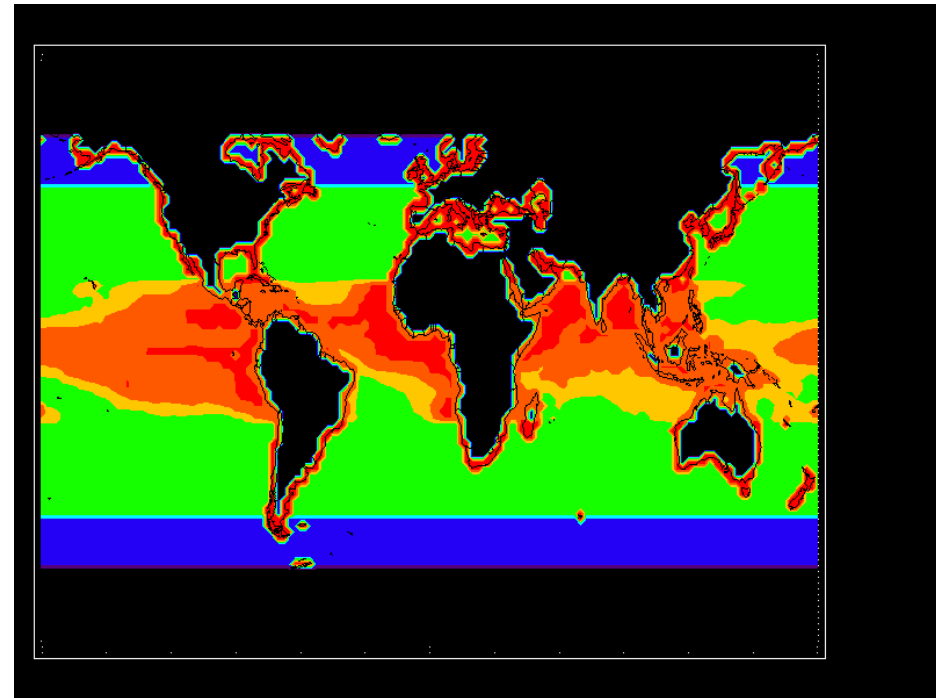
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Tropospheric Halogen Chemistry

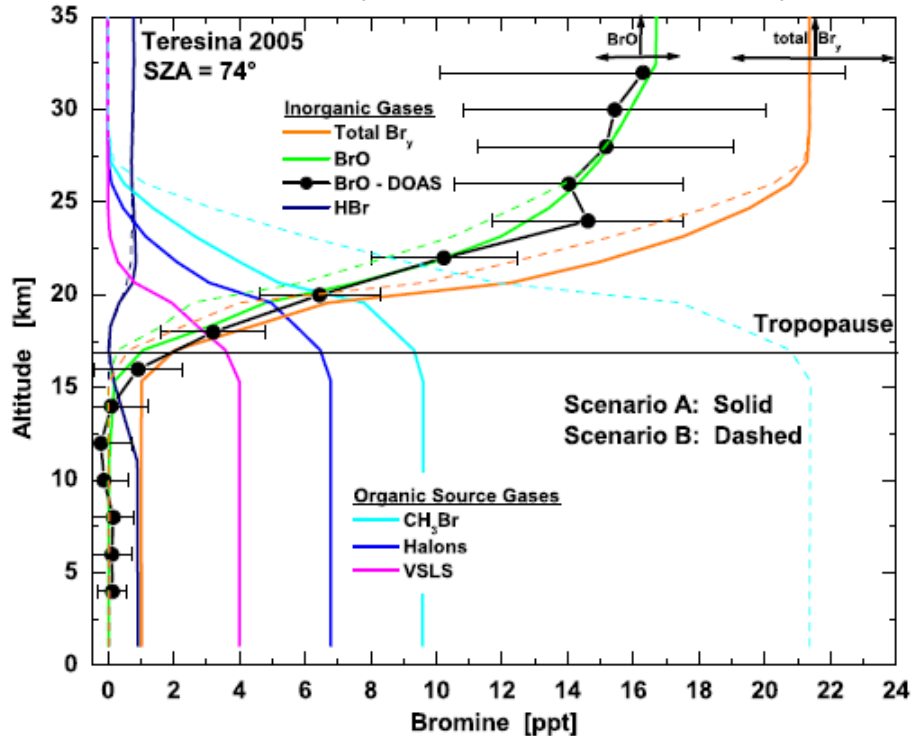
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VSL halogen sources in CAM-Chem



Ordoñez et al., ACP, 2012
Saiz-Lopez et al., ACP, 2012

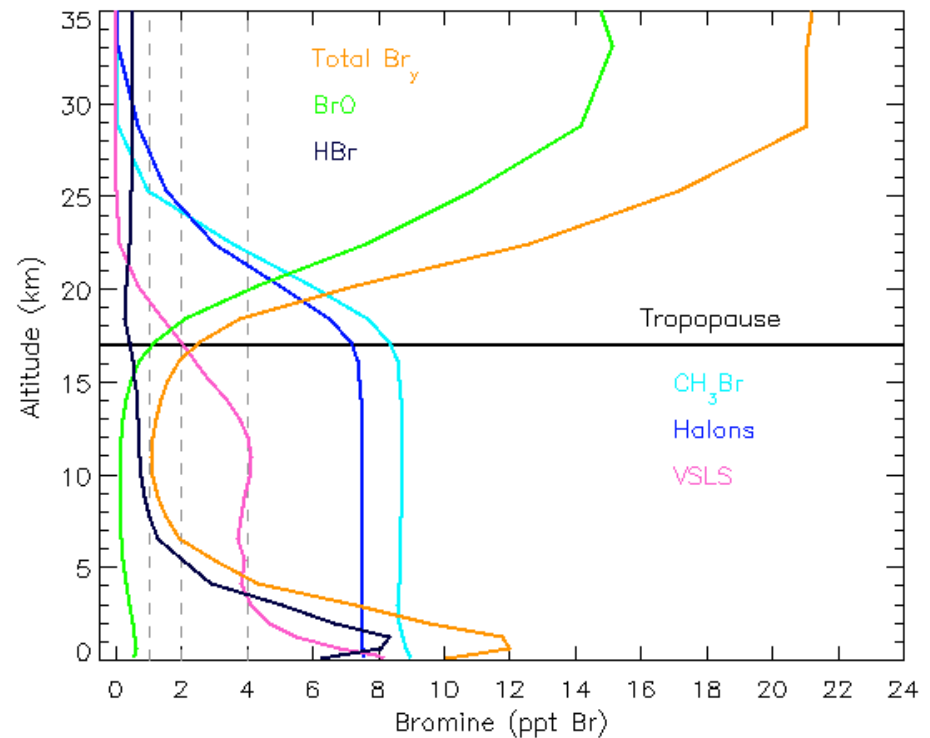
Teresina (Dorf et al., ACP, 2008)



Notes:

- SLIMCAT run with CH₃Br (9.6 ppt), halons (6.8 ppt), and VLSL (4 ppt as CH₂Br₂) plus PGs (1 ppt as HBr).
- Photochemical breakdown only in stratosphere.

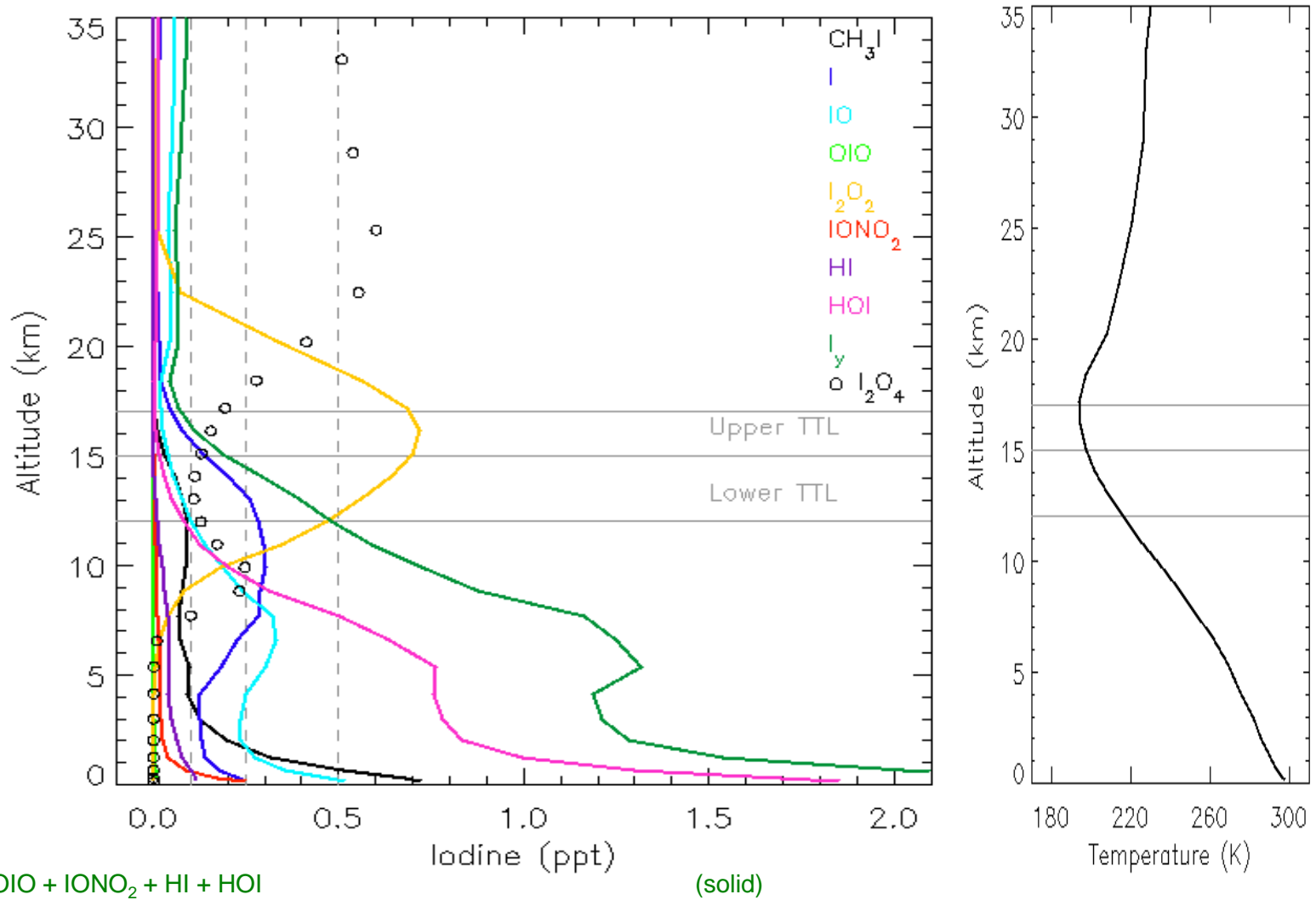
CAM-Chem_bromine (23° N – 23° S)



Notes CAM-Chem:

- Halons = H-1211 + H-1301 (i.e. CF₂CLBR + CF₃BR)
- VLSL = 3 CHBr₃ + 2 CH₂Br₂ + CH₂BrCl + 2 CHBr₂Cl + CHBrCl₂
- Total Br_y = Br + BrO + HBr + BrONO₂ + BrCl + HOBr

Daytime profiles in tropics



$$I_y = I + IO + OIO + IONO_2 + HI + HOI$$

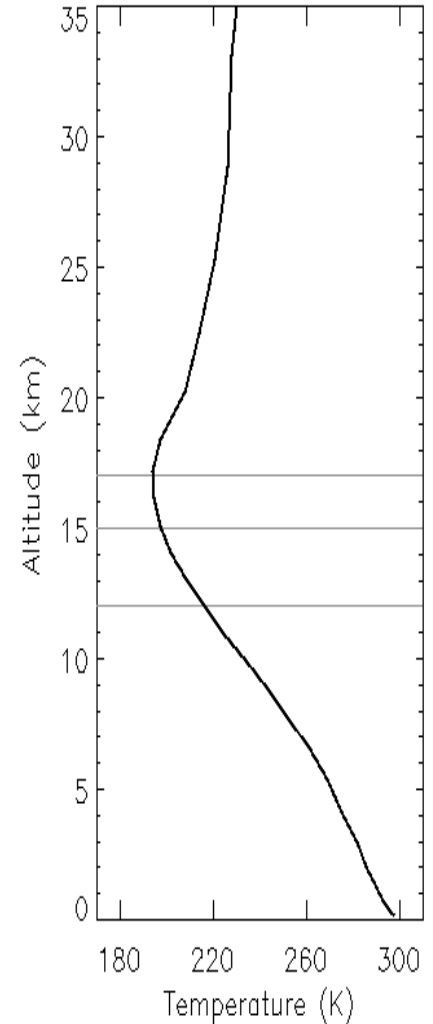
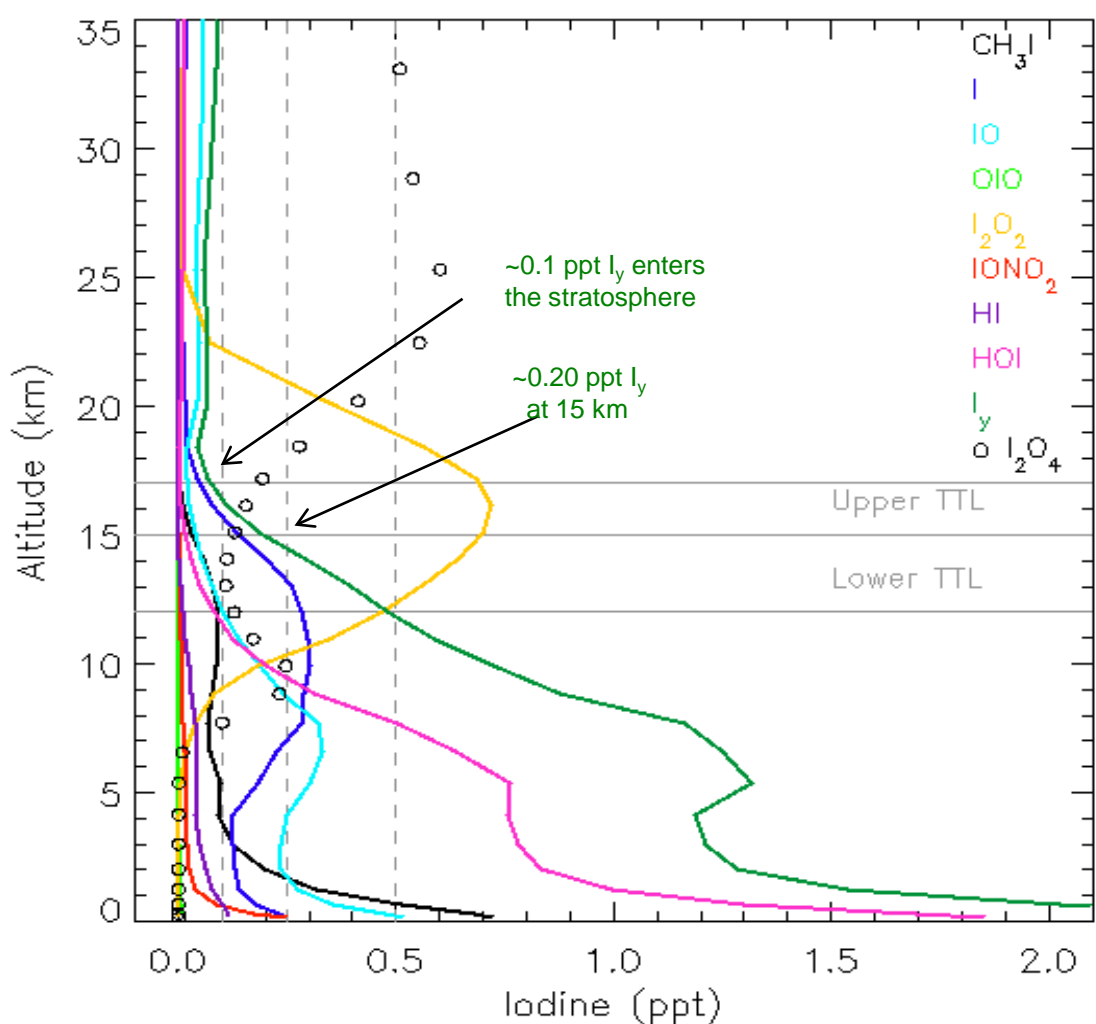
(solid)

Measured TTL to CPT abundances (Table 1-9 of WMO, 2011): :

IO, OIO PG < 0.1 ppt (Upper limits reported by Butz et al., 2009)

Table 1-7 indicates a mean of 0.12 ppt (range 0.00-0.23 ppt) CH_3I for the lower TTL (12-14 km)

Daytime profiles in tropics



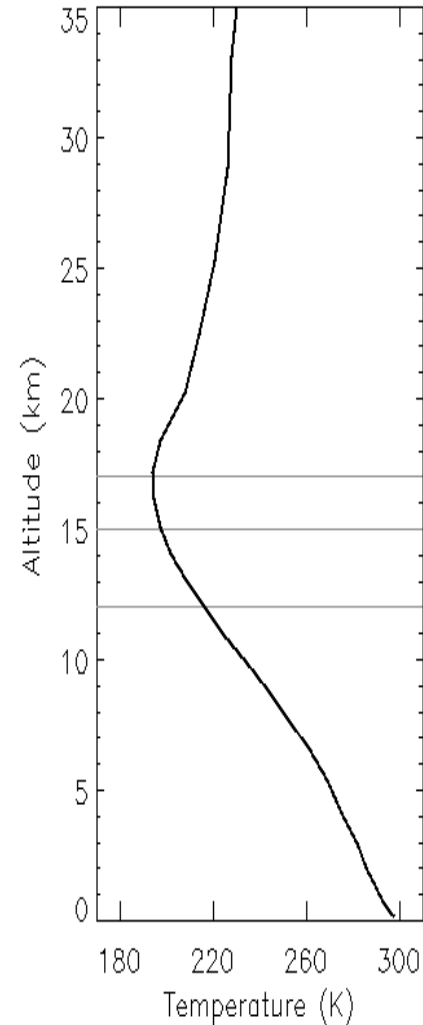
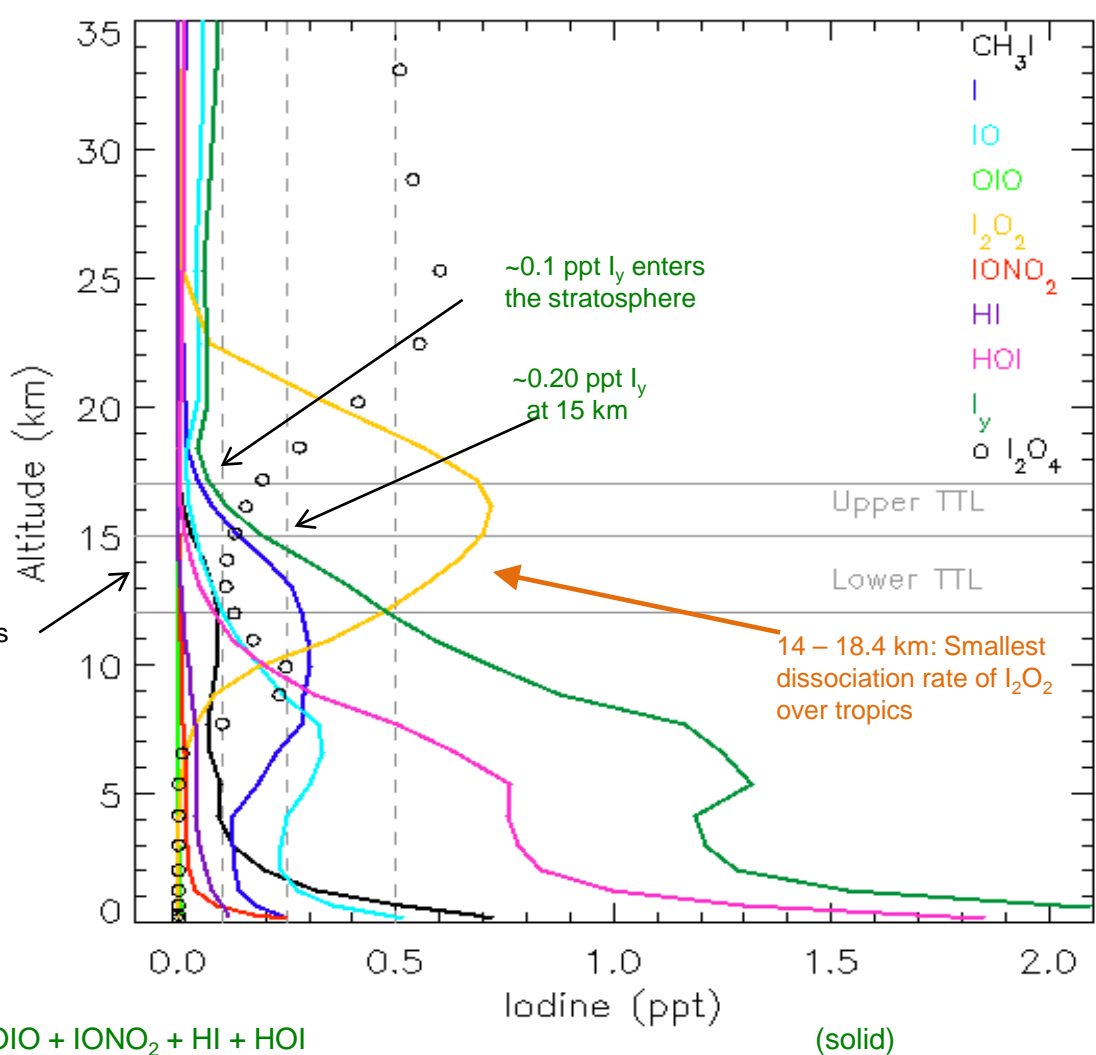
$$I_y = I + IO + OIO + IONO_2 + HI + HOI \quad (\text{solid})$$

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Daytime profiles in tropics



Only ~0.05 ppt CH_3I enters the TTL, while some obs. suggest ~0.10 ppt CH_3I

$$I_y = I + IO + OIO + IONO_2 + HI + HOI$$

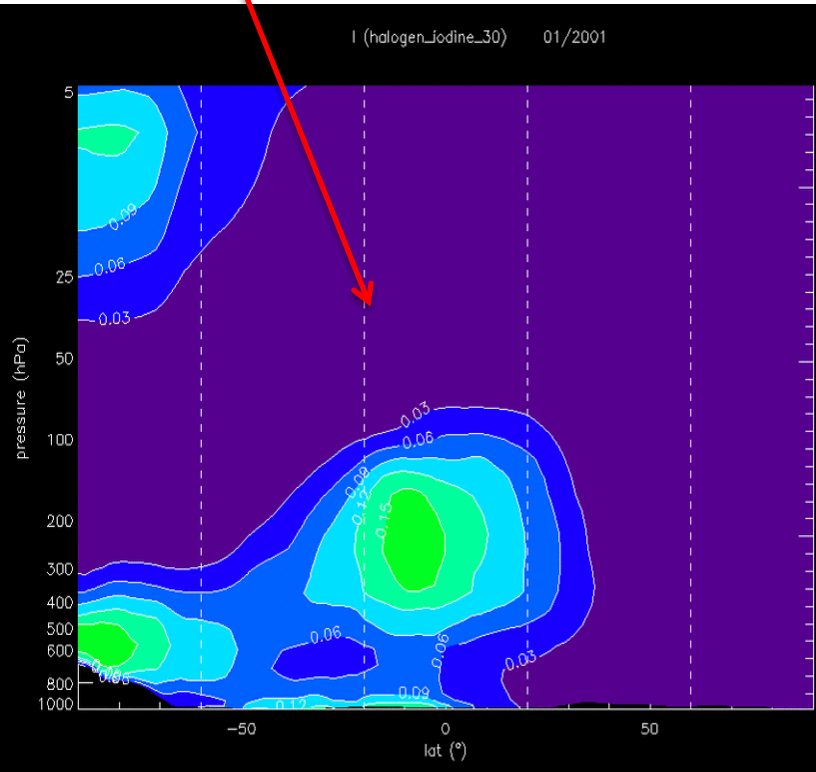
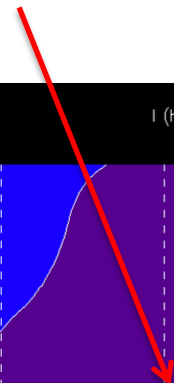
(solid)

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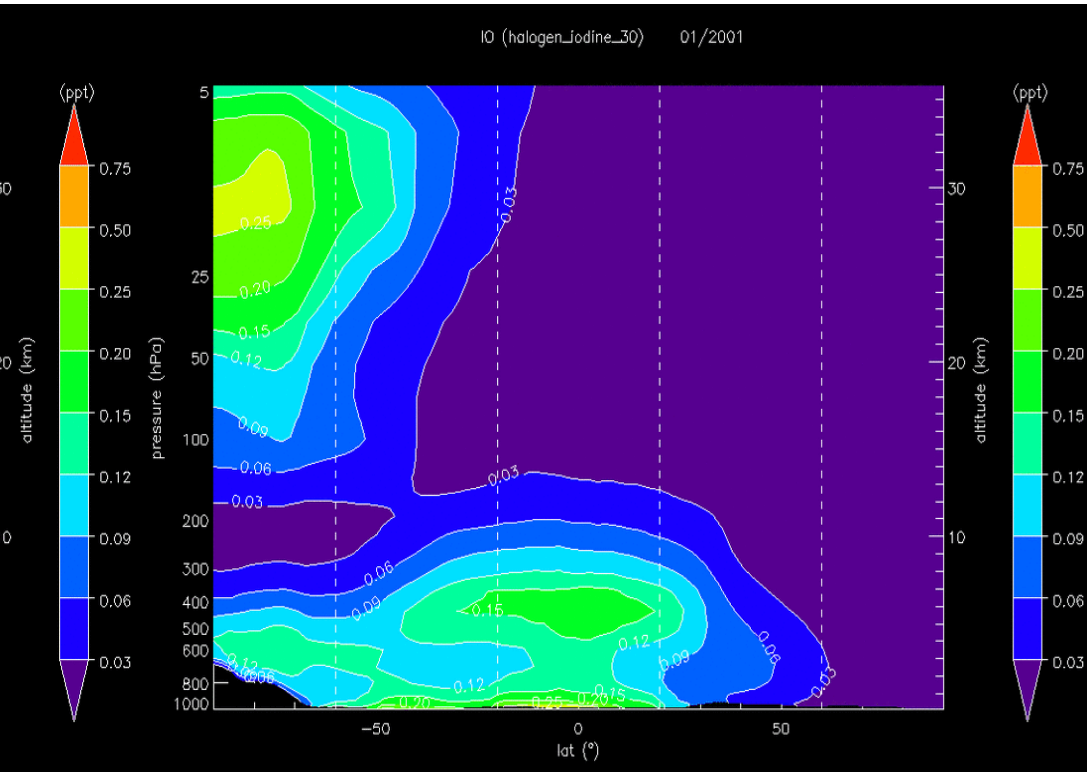
Zonal averages of atomic iodine and IO:

- Largest loading of IO_x is not in the MBL
- Height shift in the I and IO concentration peaks

Tropical "iodine mushroom cloud"



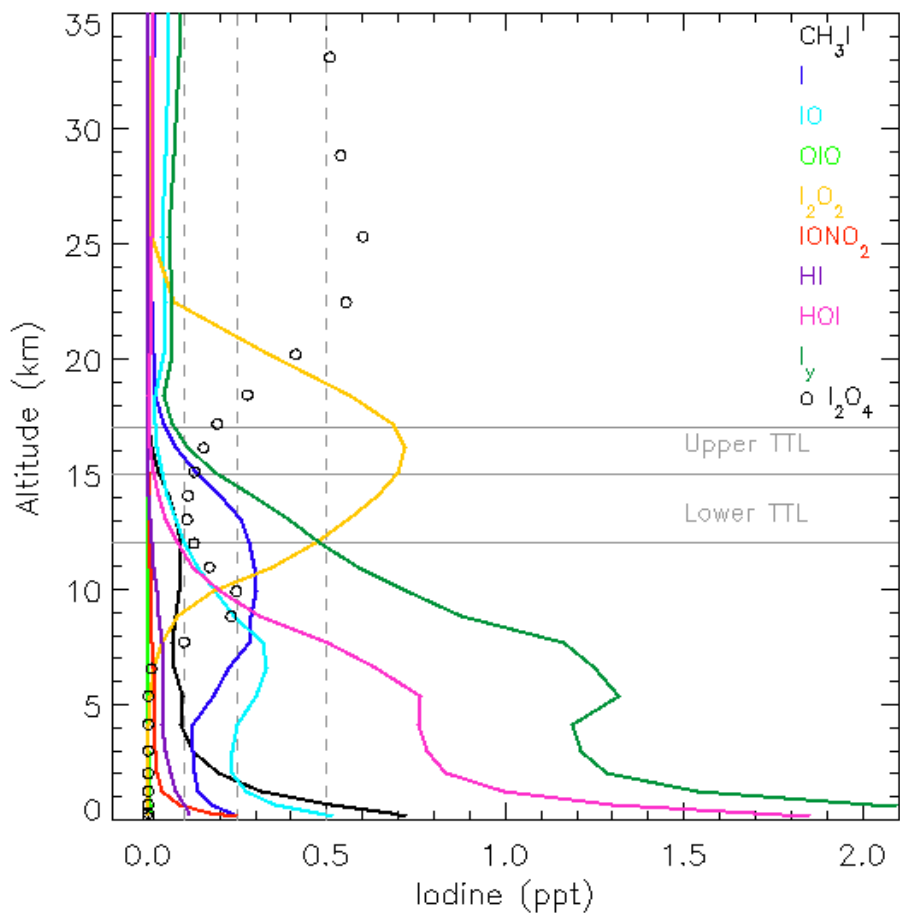
Peak I height: 10-12 km



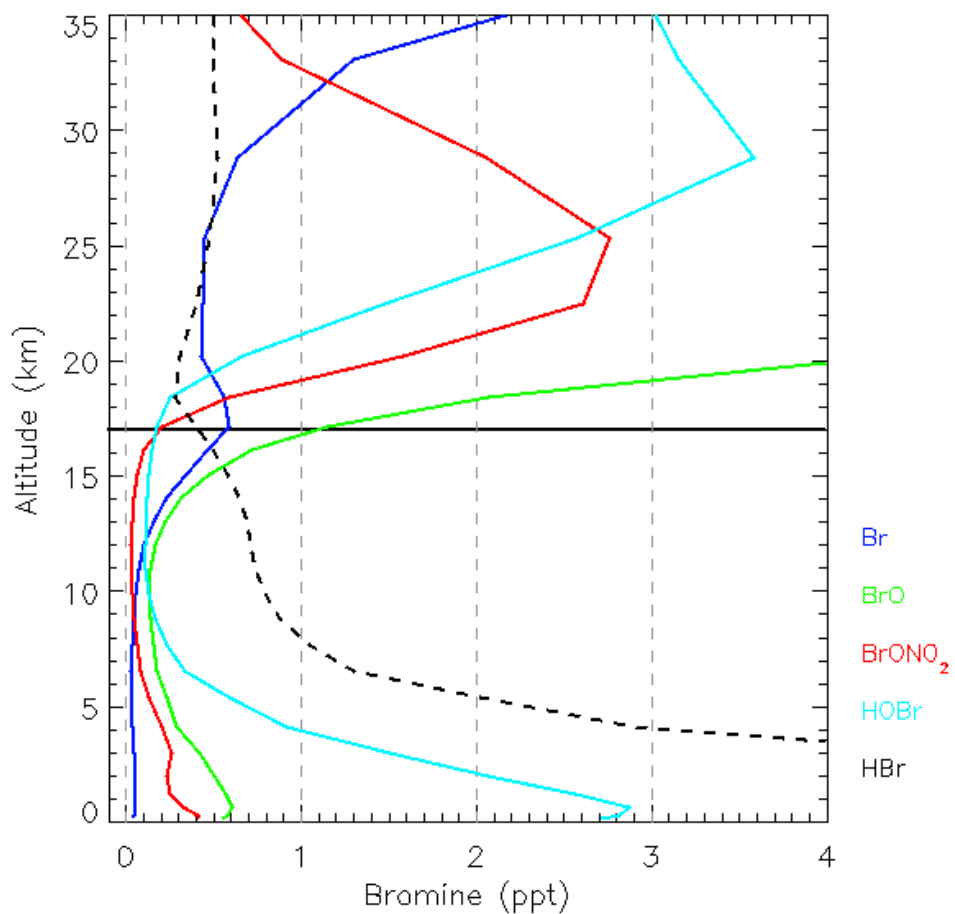
Peak IO height: 5-8 km

Annually average profiles in tropics:

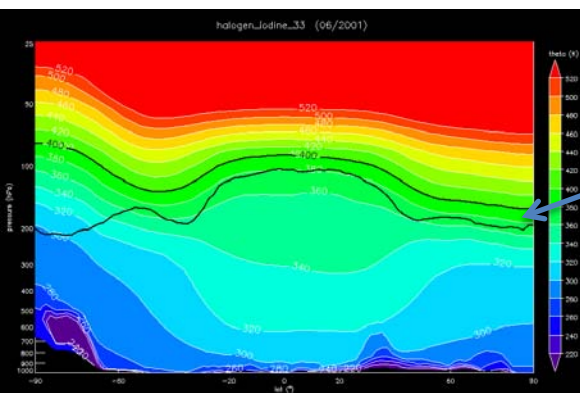
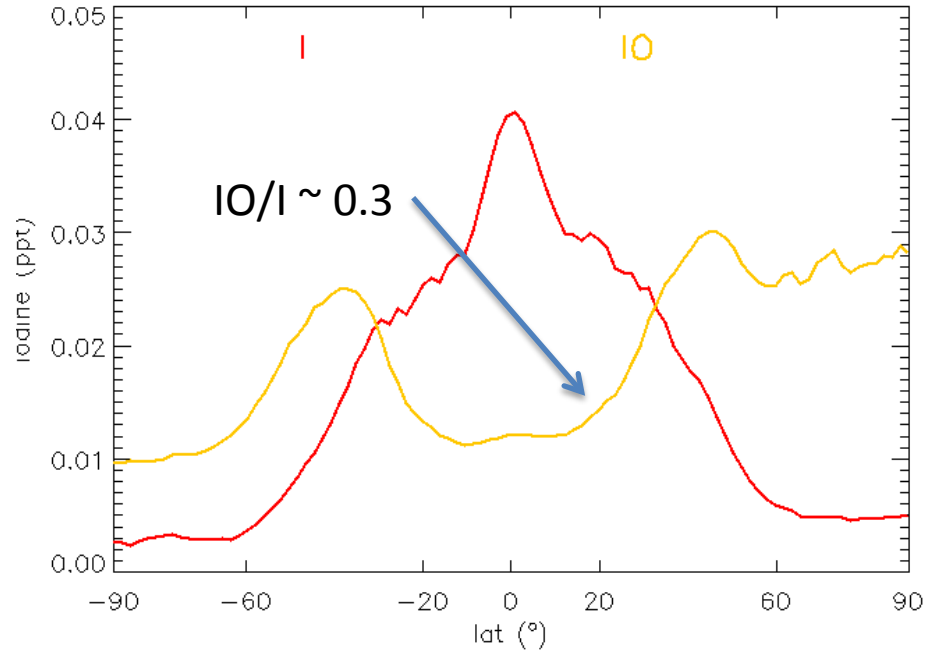
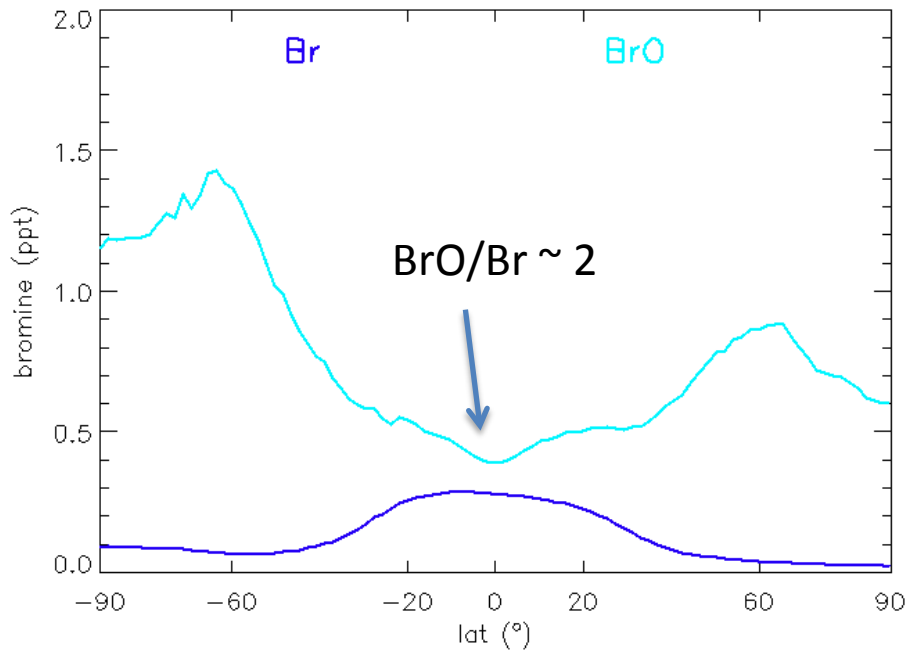
Iodine species

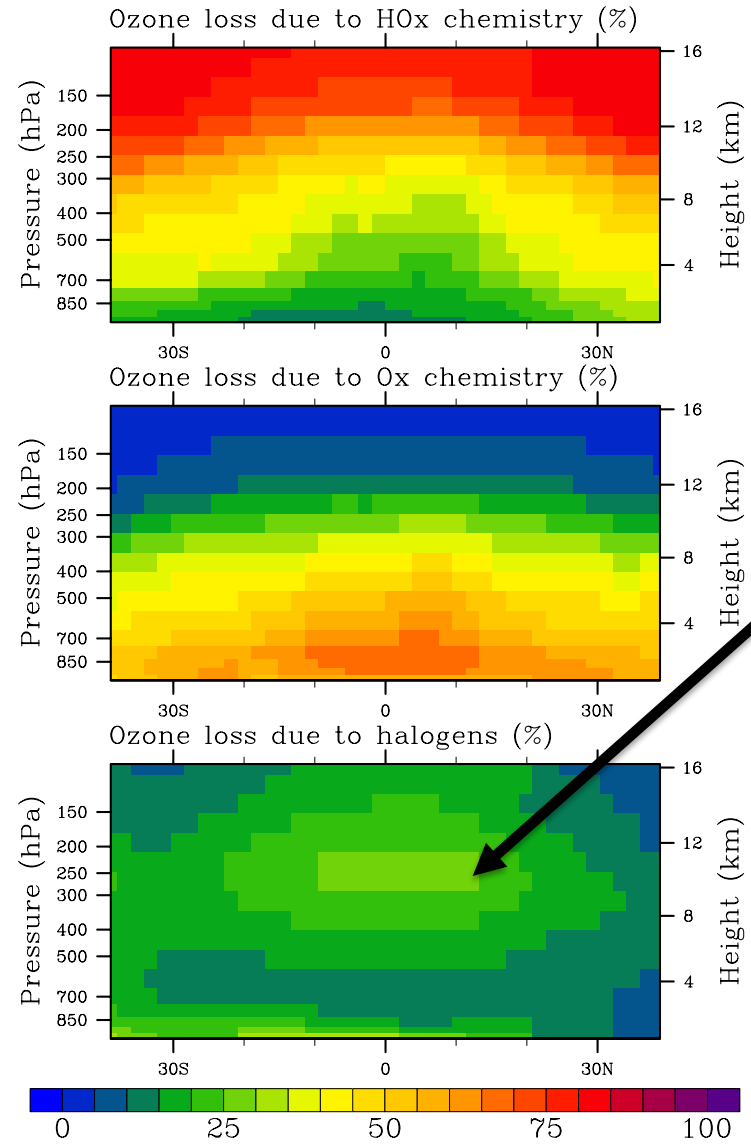


Bromine species



Annually average of latitudinal dependence of X/XO in LMS:

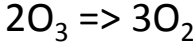
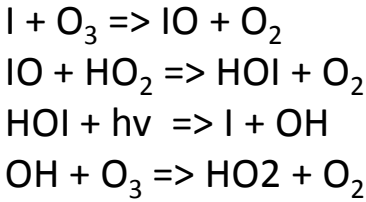
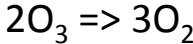
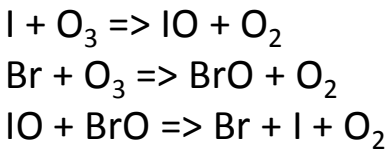




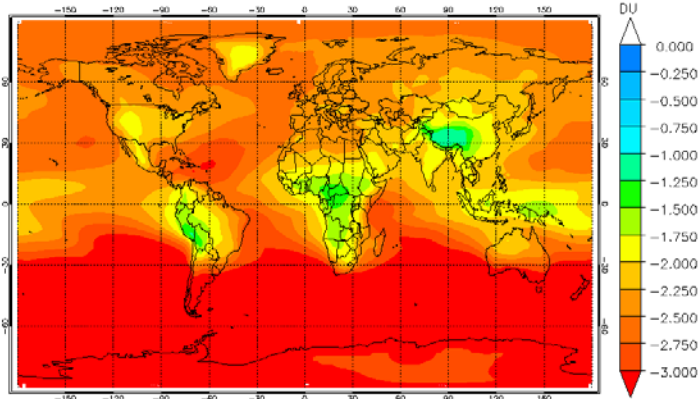
Zonal annual mean ozone loss fraction (due to chemistry)

Up to ~30% in tropical UT

Iodine Chemistry:



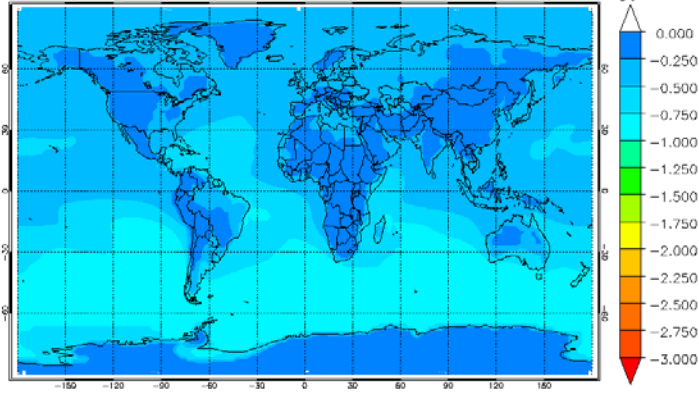
O₃ column difference (200 hPa – surface) (DU) (halogens – no halogens)



Troposphere (200 hPa – surface);

max	mean	min
-4.8	-2.8	-0.9

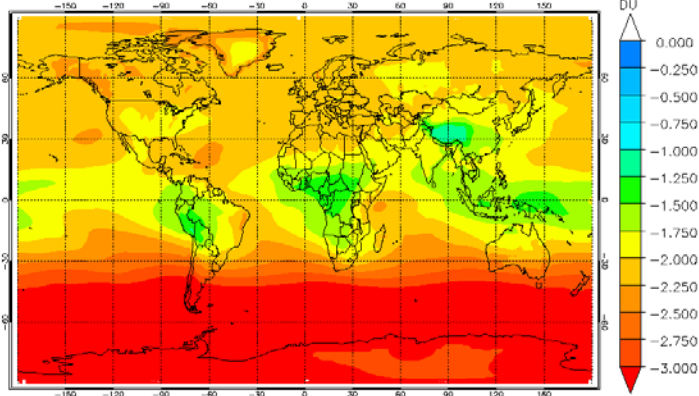
↓
~ 11% of trop. column



BL (850 hPa – surface):

max	mean	min
-0.9	-0.4	-0.1

15%



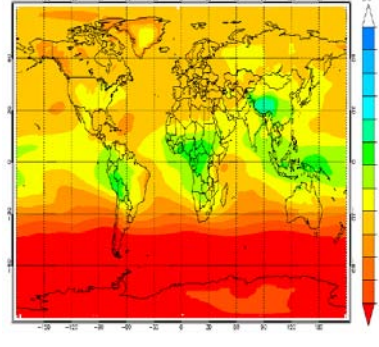
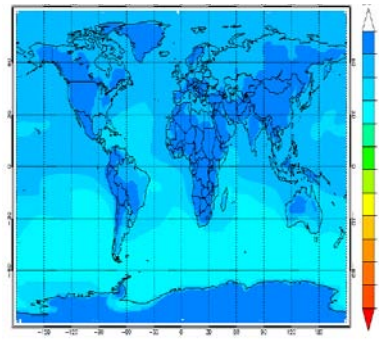
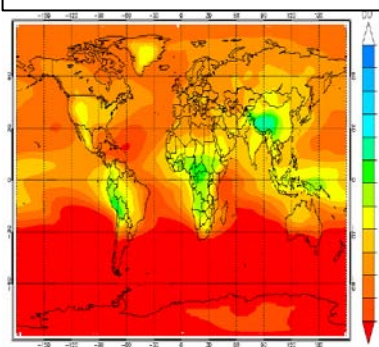
FT (850 hPa – 200 hPa):

max	mean	min
-4.3	-2.4	-0.8

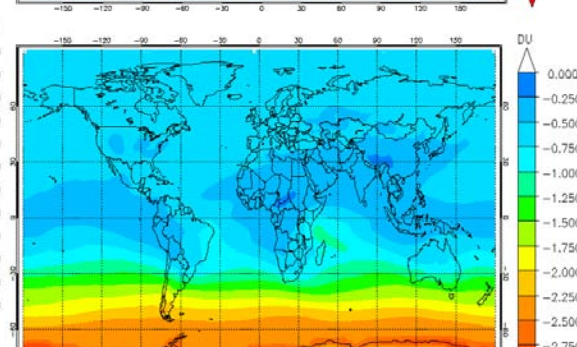
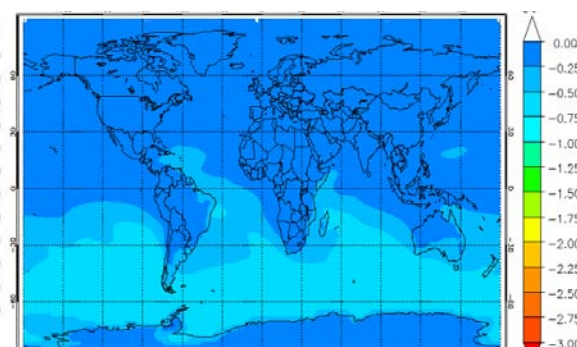
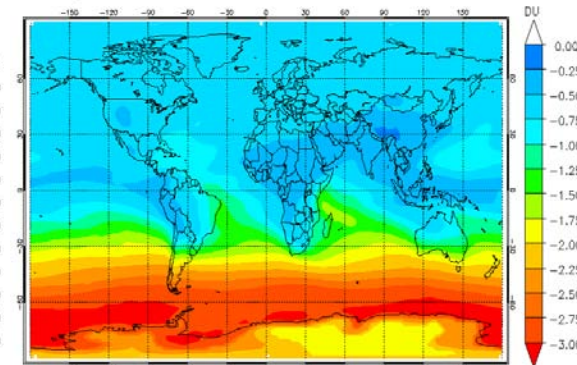
85%

Note:
no ice emissions included.

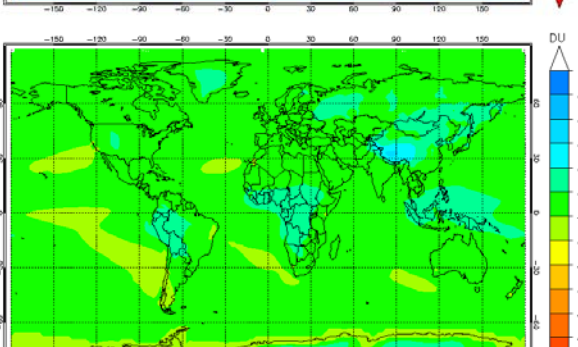
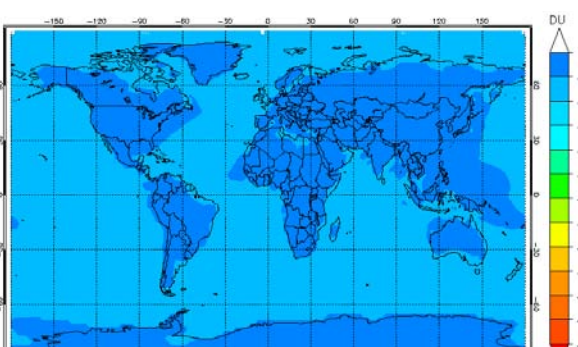
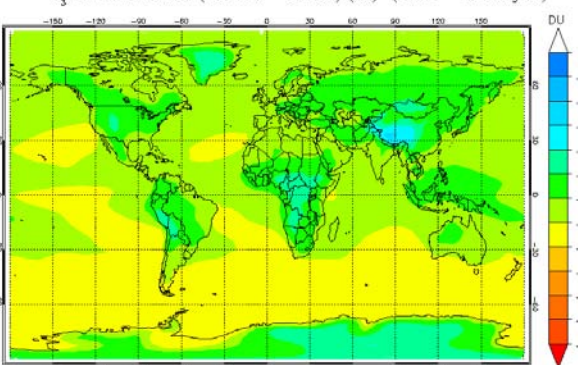
Cl + Br + I



Bromine



Iodine



Tot
40%

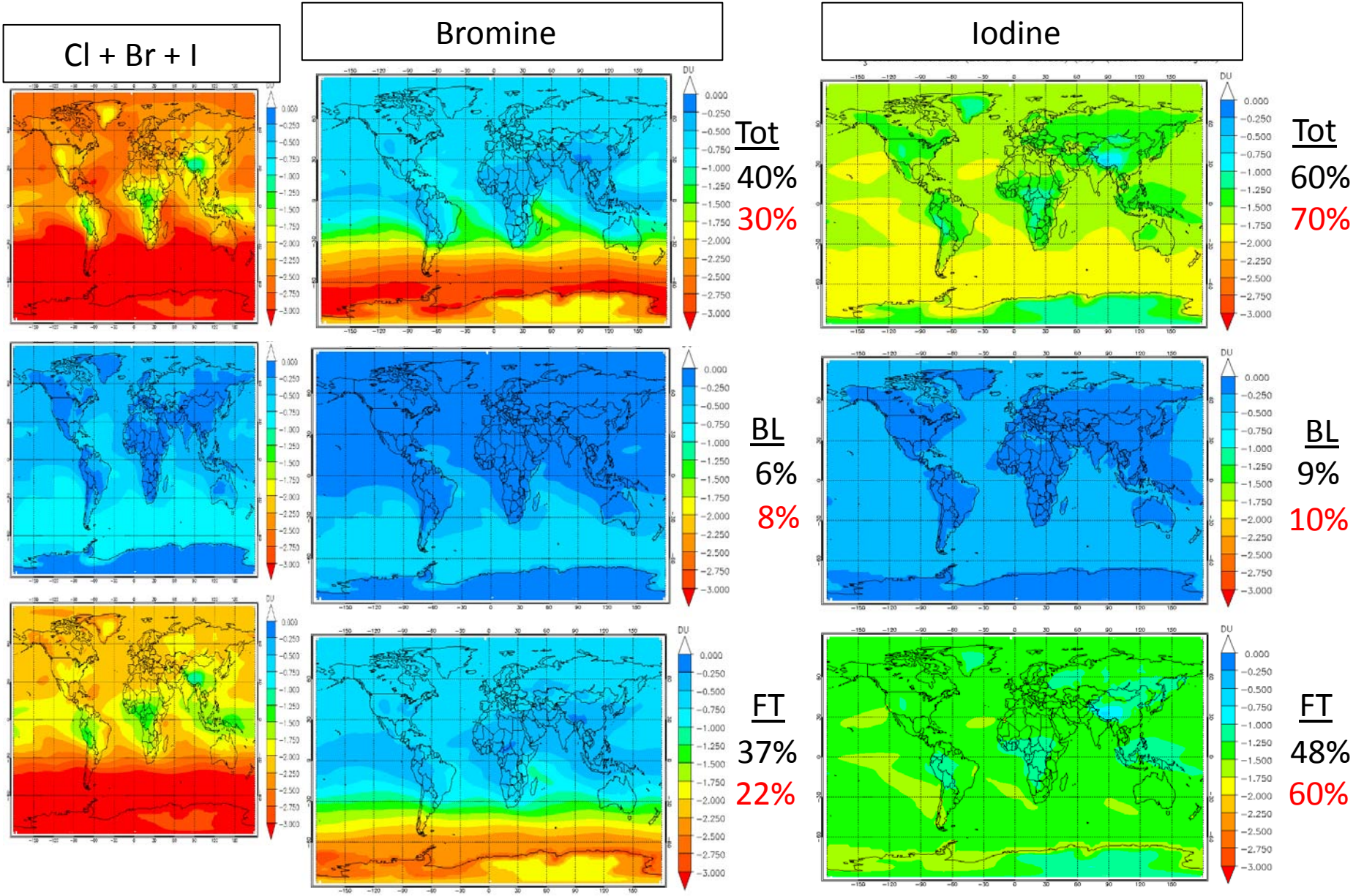
BL
6%

FT
37%

Tot
60%

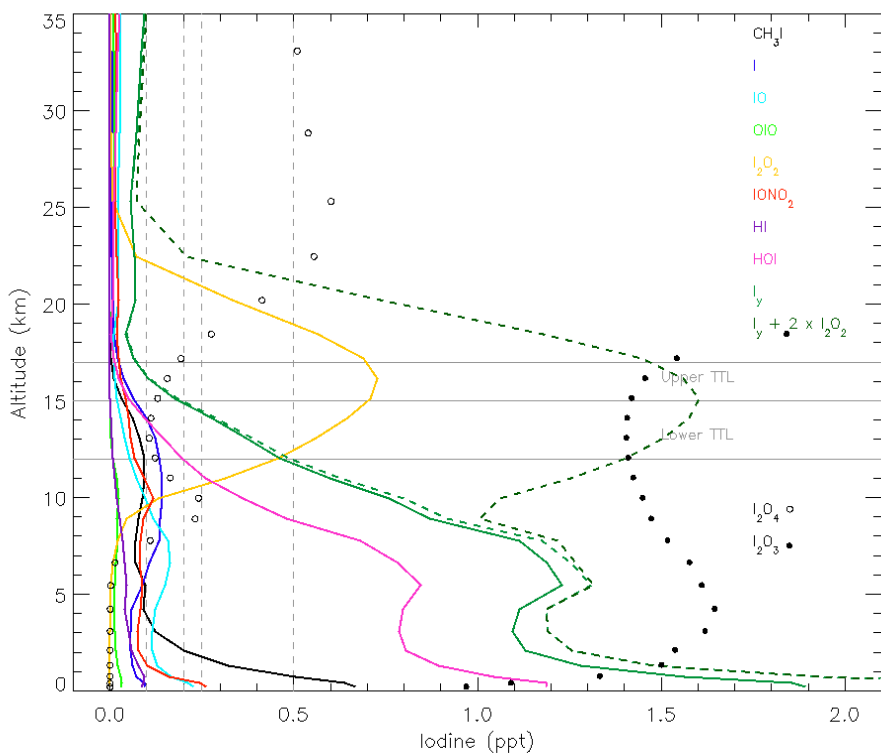
BL
9%

FT
48%

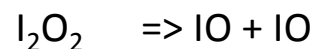
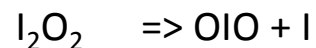
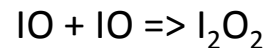


High levels of I_2O_2 , I_2O_3 , I_2O_4

halogen_iodine_30 (01-12/2001)



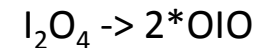
I_2O_2 Chemistry in the model...



I_2O_3 Chemistry in the model...



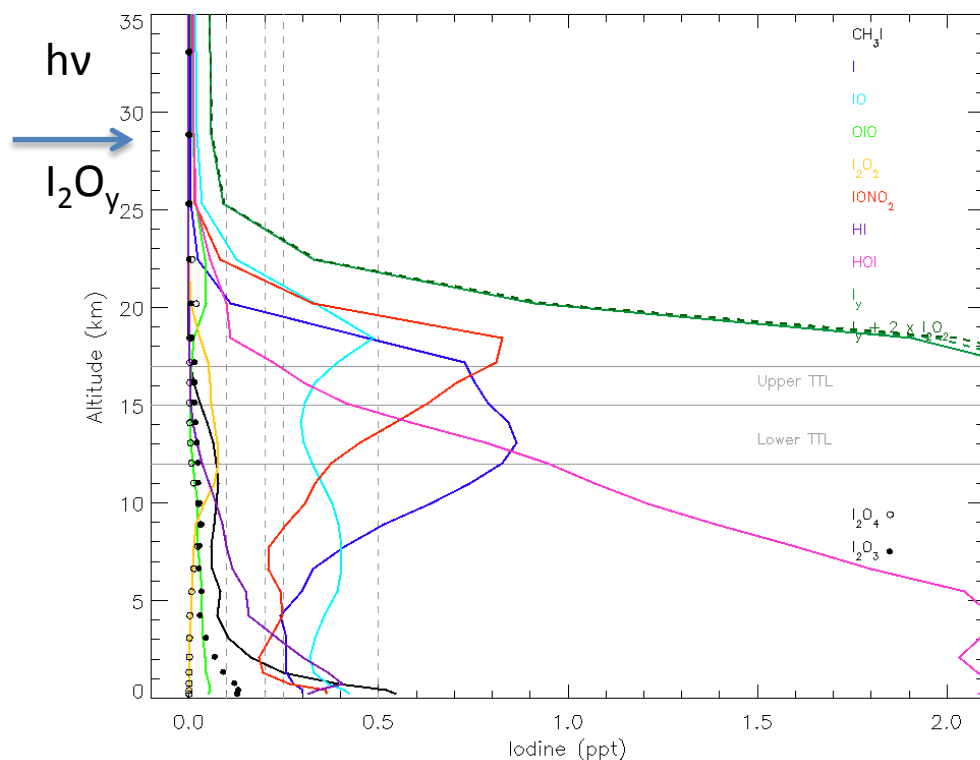
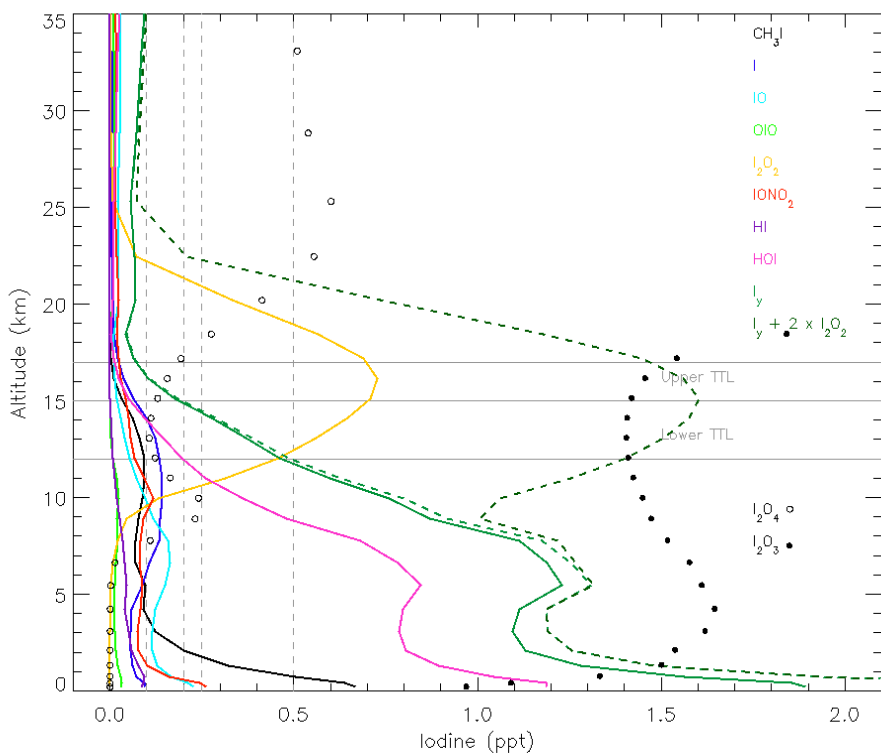
I_2O_4 Chemistry in the model...



High levels of I_2O_2 , I_2O_3 , I_2O_4

halogen_iodine_30 (01-12/2001)

halogen_iodine_35 (05/2000)

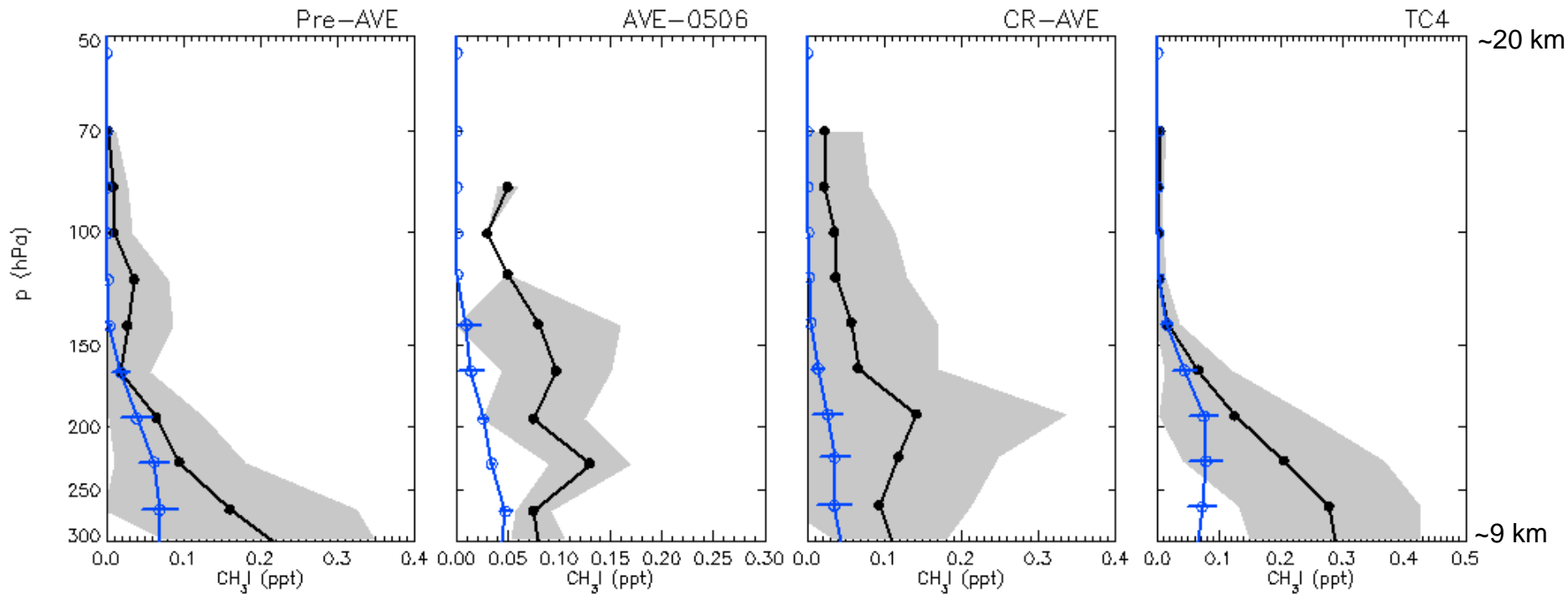


Ozone impact:

Ozone depletion efficiency by iodine is enhanced 2-3 times if I_2O_y photolysis is included !

- Oceanic sources and chemistry of **iodine** implemented in a **CCM**.
- Iodine partitioning shows high I/IO ratio in tropical UTLS
- Iodine-mediated ozone depletion, compared to bromine, is **predominant** throughout most of the troposphere and significant in the tropical LMS
- Experimental work on I_2O_y is key to further determine the role of iodine in UTLS O₃ destruction

Methyl iodide (CH₃I)



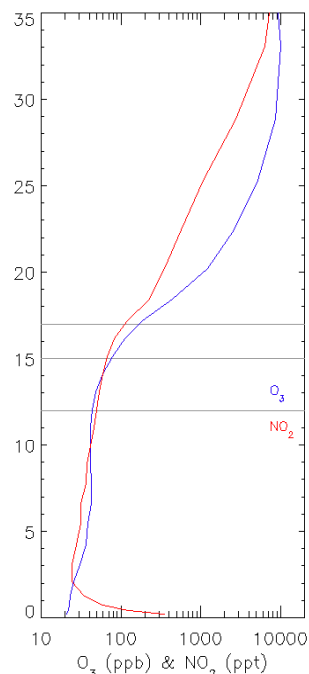
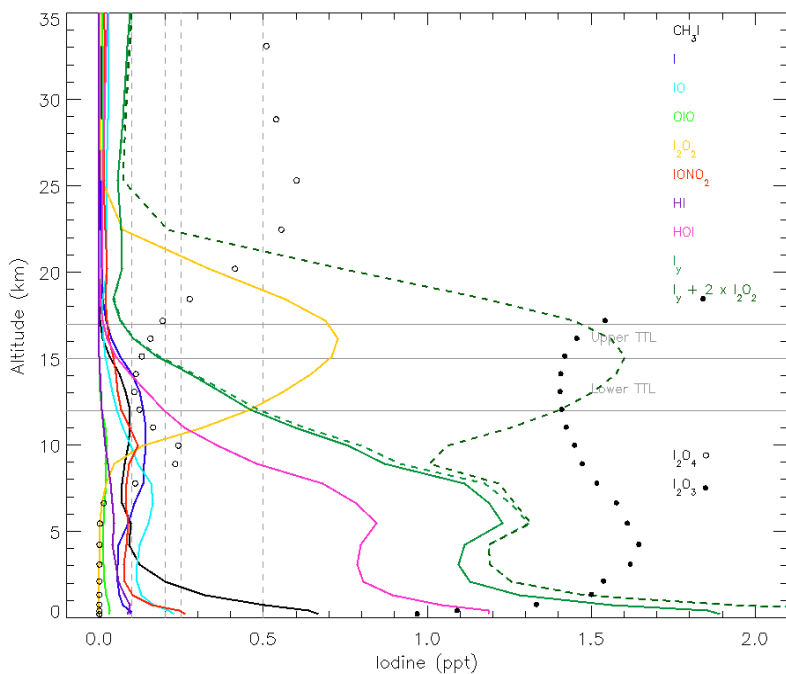
aircraft (mean ± stdev)

model (mean ± stdev)

Underestimation of CH₃I, and possibly of
O₃ loss by iodine chemistry in the UTLS

High levels of I_2O_2 , I_2O_3 , I_2O_4

halogen_iodine_30 (01-12/2001)



Estimation of XS of I_2O_y

